

Solution Stoichiometry



$$\frac{0.29 \text{ mol KI}}{\cancel{L}} \times \frac{0.037 \cancel{L}}{1} \times \frac{1 \text{ mol Pb}(\text{NO}_3)_2}{2 \text{ mol KI}} \times \frac{1}{0.042 \text{ L}}$$

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$$M = \frac{\text{mol}}{\text{L}}$$

Chemistry Solution Stoichiometry

Charles E. Mortimer



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