

Chapter 4 Test, Form 2A

SCORE _____

Write the letter for the correct answer in the blank at the right of each question.

1. What is the slope-intercept form of the equation of a line with a slope of 5 and a y -intercept of -8 ?

A $y = -8x + 5$ B $y = 8x - 5$ C $5x - y = -8$ D $y = 5x - 8$

1. _____

2. Which equation is graphed at the right?

F $2y - x = 10$ H $2x - y = 5$
 G $2x + y = -5$ J $2y + x = -5$



2. _____

3. Which is an equation of the line that passes through $(2, -5)$ and $(6, 3)$?

A $y = \frac{3}{2}x - 6$ C $y = 2x + 12$

B $y = \frac{1}{2}x$ D $y = 2x - 9$

3. _____

4. What is an equation of the line through $(0, -3)$ with slope $\frac{2}{5}$?

F $-5x + 2y = 15$ H $2x - 5y = 15$
 G $-5x - 2y = -15$ J $-2x + 5y = 15$

4. _____

5. Which is an equation of the line with slope -3 and a y -intercept of 5 ?

A $y = -3(x + 5)$ B $y - 5 = -3x$ C $-3x + y = 5$ D $y = 5x - 3$

5. _____

6. What is the equation of the line through $(-2, -3)$ with a slope of 0 ?

F $x = -2$ G $y = -3$ H $-2x - 3y = 0$ J $-3x + 2y = 0$

6. _____

7. Find the slope-intercept form of the equation of the line that passes through $(-5, 3)$ and is parallel to $12x - 3y = 10$.

A $y = -4x - 17$ B $y = 4x - 13$ C $y = -4x + 13$ D $y = 4x + 23$

7. _____

8. If line q has a slope of $-\frac{2}{8}$, what is the slope of any line perpendicular to q ?

F $-\frac{2}{8}$ G $\frac{2}{8}$ H $\frac{8}{2}$ J $-\frac{8}{2}$

8. _____

9. A line of fit might be defined as

A a line that connects all the data points.
 B a line that might best estimate the data and be used for predicting values.
 C a vertical line halfway through the data.
 D a line that has a slope greater than 1.

9. _____

10. A scatter plot of data comparing the number of years since Holbrook High School introduced a math club and the number of students participating contains the ordered pairs $(3, 19)$ and $(8, 42)$. Which is the slope-intercept form of an equation for the line of fit containing those points?

F $y = 4.6x + 5.2$ G $y = 3x + 1$
 H $y = 5.2x + 4.6$ J $y = 0.22x - 1.13$

10. _____

11. Use the equation from Question 10 to estimate the number of students who will be in the math club during the 15th year.

A 53 B 61 C 65 D 74

11. _____

Chapter 4 Test Form 2a Answer Key

Yunker



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The Hospital Corps Quarterly United States. Navy Department. Bureau of Medicine and Surgery, 1945

Chemical Interactions, 2005

Holt Algebra 1 2003 Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003

A Teachers' Manual in the Use of the Educational Scales Marvin James Van Wagenen, 1928

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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g})$ $\Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g})$ $\Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g})$ $\Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g})$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l})$ $\Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l})$ $\Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta h = 285 \text{ kJ/mole}$

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web thus for reaction 9 2 in which 1 mole of h₃o and 1 mole of oh combine to form 2 moles of h₂o h 55 8 kj as shown note that the enthalpy change h hproducts hreactants is positive if heat is absorbed that is if hproducts

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web hess s law enthalpy of reaction and hess s law abstract the goal of this lab is to perform experiments and practice hess s law by determining enthalpy change in chemical reactions in this lab three experiments were conducted in order to test hess s law the experiments required observing and collecting data on three different reactions

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web objective the purpose of the lab is to combine two reactants in the calorimeter and measure the heats of reaction in solution eventually allowing an individual to prove or disprove hess s law introduction thermodynamics is the study of energy exchanged between the system and the surroundings

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web 3 12 hess s law in this experiment you will measure the amount of heat released in these three related exothermic reactions 1 naoh s na aq oh aq h 1 2 naoh s h aq cl aq h 2 o na aq cl aq h 2 3

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web tell the students why it has special significance for you note this works better if there is an anecdote or story attached to your object now put the students into groups in breakout rooms and ask them to show and talk about their objects when they have finished you can ask students which was the most interesting object they saw show

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