

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. c In a Stock name such as iron(III) sulfate, the roman numeral tells us _____
- how many atoms of Fe are in one formula unit
 - how many sulfate ions can be attached to the iron atom
 - the charge on each Fe ion
 - the total positive charge of the formula unit
2. c The result of changing a subscript in a correctly written chemical formula is to _____
- change the number of moles represented by the formula
 - change the charges on the other ions in the compound
 - change the formula so that it no longer represents the compound it previously represented
 - have no effect on the formula
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
- | | |
|--|---|
| <u> 4 elements </u> | a. How many elements make up this compound? |
| <u> 6 oxygen atoms </u> | b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$? |
| <u> 21 atoms </u> | c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$? |
| <u> 4.2×10^{24} atoms </u> | d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$? |
4. How many atoms are present in each of these formula units?
- | | |
|---------------------|-------------------------|
| <u> 11 atoms </u> | a. $Ca(HCO_3)_2$ |
| <u> 45 atoms </u> | b. $C_{12}H_{22}O_{11}$ |
| <u> 10 atoms </u> | c. $Fe(ClO_2)_3$ |
| <u> 9 atoms </u> | d. $Fe(ClO_3)_2$ |
5. N_2O_5 a. What is the formula for the compound dinitrogen pentoxide?
- carbon(IV) sulfide b. What is the Stock name for the covalent compound CS_2 ?
- H_2SO_3 c. What is the formula for sulfurous acid?
- phosphoric acid d. What is the name for the acid H_3PO_4 ?

Chapter 7 Chemical Formulas And Compounds Test B

J Ma



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Table of Contents Chapter 7 Chemical Formulas And Compounds Test B

1. Understanding the eBook Chapter 7 Chemical Formulas And Compounds Test B
 - The Rise of Digital Reading Chapter 7 Chemical Formulas And Compounds Test B
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 7 Chemical Formulas And Compounds Test B
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 7 Chemical Formulas And Compounds Test B
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 7 Chemical Formulas And Compounds Test B
 - Personalized Recommendations
 - Chapter 7 Chemical Formulas And Compounds Test B User Reviews and Ratings
 - Chapter 7 Chemical Formulas And Compounds Test B and Bestseller Lists

5. Accessing Chapter 7 Chemical Formulas And Compounds Test B Free and Paid eBooks
 - Chapter 7 Chemical Formulas And Compounds Test B Public Domain eBooks
 - Chapter 7 Chemical Formulas And Compounds Test B eBook Subscription Services
 - Chapter 7 Chemical Formulas And Compounds Test B Budget-Friendly Options
6. Navigating Chapter 7 Chemical Formulas And Compounds Test B eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 7 Chemical Formulas And Compounds Test B Compatibility with Devices
 - Chapter 7 Chemical Formulas And Compounds Test B Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 7 Chemical Formulas And Compounds Test B
 - Highlighting and Note-Taking Chapter 7 Chemical Formulas And Compounds Test B
 - Interactive Elements Chapter 7 Chemical Formulas And Compounds Test B
8. Staying Engaged with Chapter 7 Chemical Formulas And Compounds Test B
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 7 Chemical Formulas And Compounds Test B
9. Balancing eBooks and Physical Books Chapter 7 Chemical Formulas And Compounds Test B
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 7 Chemical Formulas And Compounds Test B
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 7 Chemical Formulas And Compounds Test B
 - Setting Reading Goals Chapter 7 Chemical Formulas And Compounds Test B
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 7 Chemical Formulas And Compounds Test B
 - Fact-Checking eBook Content of Chapter 7 Chemical Formulas And Compounds Test B
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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