

Chapter Test B

Chapter Chemical Equations and Reactions

Write the full name (including units) for the term at the end of the sentence. Please show your work. Please write answers in full sentences with units.

- The molar volume of an ideal gas at standard temperature and pressure is 22.7 L mol^{-1} .
 - gas
 - pressure
 - standard temperature
 - volume
- The volume of a gas is independent of its pressure at constant temperature.
 - constant
 - volume
 - standard temperature
 - number of particles
- An unknown substance has a molar mass of 100 g mol^{-1} .
 - mass
 - density of water
 - volume
 - number
- The molar mass of a compound is defined as the mass of one mole of that compound.
 - substance, the molar mass is the ratio of the mass of the substance to the number of particles
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