

WORKSHEET
CHEMISTRY FOR ENGINEERING STUDENTS, LECTURE 1
 CHEMISTRY OF METALS

ANSWERS KEY

QUESTION 1. CLASSIFY THE ELEMENTS OF THE PERIODIC TABLE INTO METALS AND NON-METALS.

1. Distinguish between metals and non-metals based on the following criteria:
 - (a) Write the balanced chemical equation for the reaction of a metal with oxygen at room temperature.

$M + O_2 \rightarrow MO$ (where M is a metal)
 - (b) Explain the difference in the physical properties of metals and non-metals.

Metals are shiny, malleable, ductile, and good conductors of heat and electricity. Non-metals are dull, brittle, and poor conductors of heat and electricity.
 - (c) Write the balanced chemical equation for the reaction of a metal with water.

$M + H_2O \rightarrow MO + H_2$ (where M is a metal)
2. State the difference between metals and non-metals based on the following criteria:
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