

Chapter 10 test**Matching**

Match each item with the correct statement below.

- | | |
|----------------------------|--------------------------------------|
| a. representative particle | d. percent composition |
| b. mole | e. standard temperature and pressure |
| c. Avogadro's number | f. empirical formula |

- _____ 1. the number of representative particles of a substance present in 1 mole of that substance
- _____ 2. an atom, an ion, or a molecule, depending upon the way a substance commonly exists
- _____ 3. the SI unit used to measure amount of substance
- _____ 4. 0°C and 1 atm
- _____ 5. the percent by mass of each element in a compound
- _____ 6. the smallest whole number ratio of the atoms in a compound

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- _____ 7. What SI unit is used to measure the number of representative particles in a substance?
- | | |
|-------------|-----------|
| a. kilogram | c. kelvin |
| b. ampere | d. mole |
- _____ 8. How many hydrogen atoms are in 5 molecules of isopropyl alcohol, C_3H_7O ?
- | | |
|-------------------------------------|--------------------------------------|
| a. $5 \times (6.02 \times 10^{23})$ | c. 35 |
| b. 5 | d. $35 \times (6.02 \times 10^{23})$ |
- _____ 9. All of the following are equal to Avogadro's number EXCEPT _____
- | |
|---|
| a. the number of atoms of bromine in 1 mol Br_2 |
| b. the number of atoms of gold in 1 mol Au |
| c. the number of molecules of nitrogen in 1 mol N_2 |
| d. the number of molecules of carbon monoxide in 1 mol CO |
- _____ 10. How many moles of tungsten atoms are in 4.8×10^{23} atoms of tungsten?
- | | |
|-------------------------------|-------------------------------|
| a. 8.0×10^{-2} moles | c. 1.3×10^{-1} moles |
| b. 8.0×10^1 moles | d. 1.3×10^{-2} moles |
- _____ 11. How many atoms are in 0.075 mol of titanium?
- | | |
|-------------------------|-------------------------|
| a. 1.2×10^{23} | c. 6.4×10^2 |
| b. 2.2×10^{24} | d. 4.5×10^{22} |
- _____ 12. How many molecules are in 2.10 mol CO_2 ?
- | | |
|------------------------------------|-------------------------------------|
| a. 2.53×10^{24} molecules | c. 3.49×10^{-24} molecules |
| b. 3.79×10^{24} molecules | d. 1.26×10^{24} molecules |

Chemistry Chapter 10 Test Answers

Albert A Gayle



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Elements of General Chemistry with Experiments John Harper Long, 1902.

Organic Chemistry Michael Smith, 2006-07-03. The Collins College Outline for Organic Chemistry begins with atomic orbitals and bonding and covers all critical topics including conformations, acids and bases, elimination reactions, addition reactions, amines, amino acids, carbohydrates, and nucleic acids, and everything in between. Completely revised and updated by Dr. Michael B. Smith. Each chapter of Organic Chemistry includes a test yourself section with answers and complete explanations at the end of each chapter. Also included are bibliographies for further reading as well as numerous graphs, charts, and examples. The Collins College Outlines are a completely revised in-depth series of study guides for all areas of study including the Humanities, Social Sciences, Mathematics, Science, Language, History, and Business. Featuring the most up-to-date information, each book is written by a seasoned professor in the field and focuses on a simplified and general overview of the subject for college students and, where appropriate, Advanced Placement students. Each Collins College Outline is fully integrated with the major curriculum for its subject and is a perfect supplement for any standard textbook.

Allied Health Exam, 1997.

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