

## Chapter Test II

### A. Matching

Match each item in Column I with the correct description in Column II. Write the letter of the correct description.

#### Column I

1. An unbalanced chemical equation for a reaction that obeys the law of conservation of matter.
2. The number of protons in the nucleus of an atom.
3. The number of electrons in a neutral atom.
4. The ratio of the actual weight of an element in a compound to its atomic weight.
5. The number of atoms of an element in a molecule of a compound.
6. The number of atoms of an element in a molecule of a compound.

#### Column II

- a. atomic weight
- b. limiting reagent
- c. molecular weight
- d. stoichiometry
- e. percent yield
- f. percent composition

### B. Multiple Choice

Choose the best answer for each question.

1. The number of protons in an atom of an element is equal to the number of
  - a. neutrons in the nucleus
  - b. electrons in the atom
  - c. neutrons in the nucleus
  - d. total mass of the atom
2. Which of the following is not a chemical change?
  - a. melting of ice
  - b. rusting of iron
  - c. burning of wood
  - d. boiling of water
3. The number of atoms of an element in a molecule of a compound is equal to the number of
  - a. molecules of the compound
  - b. atoms of the element in the compound
  - c. molecules of the element
  - d. atoms of the element

# Chemistry Chapter 12 Test B Stoichiometry

**JR Anderson**



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