

## Section One: Multiple-choice

(50 marks)

|         |       |       |       |       |
|---------|-------|-------|-------|-------|
| 1. D    | 2. C  | 3. B  | 4. C  | 5. D  |
| 6. C    | 7. C  | 8. D  | 9. B  | 10. C |
| 11. D   | 12. B | 13. B | 14. C | 15. A |
| 16. All | 17. B | 18. D | 19. A | 20. C |
| 21. C   | 22. B | 23. C | 24. D | 25. C |

## Section Two: Short answer

(70 marks)

## Question 26

(4 marks)

- (a) Some hydrochloric acid solution is mixed with solid sodium carbonate. (2 marks)

A white solid dissolves in a colourless solution, producing a colourless and odourless gas. (2)

(\*Must have two observations for both marks).

- (b) Some solid copper (II) hydroxide is mixed with a dilute nitric acid solution. (2 marks)

A blue solid dissolves in a colourless solution to produce a blue solution. (2)

(\*Must have two observations for both marks).

## Question 27

(6 Marks)

- (a) Write balanced chemical equations showing carbon dioxide reacting with water to form carbonic acid, and then the two successive ionisation reactions that carbonic acid undergo in water. (3 marks)



- (b) Identify a conjugate acid-base pair in this reaction, and explain why it is classified as a Brønsted – Lowry acid-base reaction. (3 marks)

Conjugate A/B pair =  $\text{CO}_3^{2-} / \text{HCO}_3^-$  (1) \*Also accept  $\text{HCO}_3^- / \text{CO}_3^{2-}$

This equation is classified as a Brønsted – Lowry acid-base reaction because in the forward reaction,  $\text{H}_2\text{O}$  donates a proton, thus acting as a B-L acid, (1)  
while  $\text{CO}_3^{2-}$  accepts a proton, thus acting as a B-L base. (1)

# Chemistry Semester 1 Exam Answer Key

**Linda D. Williams**



## Chemistry Semester 1 Exam Answer Key:

**Text Book of Physical Chemistry Semester-IV, Course-10 by BVR VENKATA RAO BASA (BVR SIR),2025-03-12** Text Book of Physical Chemistry Semester IV Course 10 by BVR is meant for BSc students pursuing Organic Chemistry Honors The syllabus may vary slightly from college to college The topics covered are dilute solutions electrochemistry phase rule photochemistry Practicals CST and conductometry Objective questions for entrance exams included and also video links for further knowledge included *Artificial Intelligence in Education. Posters and Late Breaking Results, Workshops and Tutorials, Industry and Innovation Tracks, Practitioners, Doctoral Consortium and Blue Sky* Andrew M. Olney,Irene-Angelica Chounta,Zitao Liu,Olga C. Santos,Ig Ibert Bittencourt,2024-07-01 This volume constitutes poster papers and late breaking results presented during the 25th International Conference on Artificial Intelligence in Education AIED 2024 which took place in Recife Brazil during July 8 12 2024 The 18 full papers and 92 short papers were carefully reviewed and selected from 200 submissions They are organized in topical sections as follows Part One Blue Sky Industry Innovation and Practitioner WideAIED and Late Breaking Results Part Two Late Breaking Results Doctoral Consortium Workshops and Tutorials **College Chemistry** Harold Goldwhite,John R. Spielman,1984 An outline of the basic concepts of chemistry includes discussions of scientific notation atomic structure chemical bonding and the periodic table *Integrating Sustainable Development into the Curriculum* Enakshi Sengupta,Patrick Blessinger,Taisir Subhi Yamin,2020-03-18 This book explores the value of institutions of higher education in leading the way on the topic of sustainability education by ensuring that it is well entrenched in the curriculum as well as everyday practice and lifestyles **Nursing School Entrance Exam Review** William H. Brown,Frederick A. Bettelheim,Jerry March,2004-04 **Tests in Chemistry** Raymond Bedell Brownlee,1937 **The Publishers Weekly** ,1967 *Reports from the Classroom* Sarah H. Huyvaert,1995

Student-Assisted Teaching Judith E. Miller,James E. Groccia,Marilyn Sue Miller,Marilyn S. Miller,2001-01-15 This book provides a range of models for undergraduate student assisted teaching partnerships to help teachers and administrators make learning more student centered effective and productive The 31 models describes a range of approaches and applications in a variety of settings and disciplines The chapters are 1 Establishing a Common Ground a Conjoint Training Model for Instructors and Peer Educators Eve M Adams Susan C Brown and Terry L Cook 2 Lessons from Peers The Design Exchange Mark J Chidister Frank H Bell Jr And Kurt M Earnest 3 Peer Teaching in the Experimental College Robyn Gittleman and Howard Woolf 4 Peer Facilitators as Lead Freshman Seminar Instructors Jean M Henscheid 5 The Teaching Teams Program a Just in Time Model for Peer Assistance Harold P Larson Reed Mencke Stacy J Tollefson Elizabeth Harrison and Elena Merman 6 The Teaching Teams Program Transforming the Role of the Graduate Teaching Assistant David A Wood Jr Jennifer L Hart Stacy J Tollefson Dawn E DeToro and Julie Libarkin 7 The Teaching Teams Program Empowering Undergraduates in a Student Centered Research University Lacey A Stover Kirstin A Story Amanda M Skousen Cynthia E

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