

Chemistry pH and pOH Worksheet

1. Calculate the values of both pH and pOH of the following solutions.

	pH	pOH
0.020 M HCl		
0.0050 M NaOH		
A blood sample 7.2×10^{-8} M of H^+		
0.0050 M NaOH		

2. Find the values of $[H^+]$, pOH, and $[OH^-]$ that correspond to each of the following pH values.

	[H ⁺]	[OH ⁻]	pOH
pH of lemon juice = 2.9			
pH of sauerkraut = 3.85			
pH of milk of magnesia = 10.81			
pH of orange juice = 4.11			
pH of diluted household ammonia = 11.61			

3. Determine which of the solutions in problem #2 are acidic.

4. A certain brand of root beer has a hydrogen concentration equal to 1.9×10^{-5} M. What are the pH and pOH of this root beer?

5. Dr. Pepper has a $[H^+] = 1.4 \times 10^{-5}$ M. What is its pH?

Chemistry If8766 Answers Ph And Poh

TD Snyder



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