
Fundamental of Electrical Engineering

Learning outcome

- To enable students to understand the basic electromagnetic phenomenon, circuit variables and parameters
- To enable students to understand and apply the fundamental and derived circuit laws and theorems to the analysis of dc circuits
- To enable students to understand and apply the fundamental and derived circuit laws and theorems to the analysis of steady state poly-phase ac circuits;

MODULE ONE

REVIEW OF ELECTROMAGNETIC PHENOMENON, VARIABLES AND CIRCUIT PARAMETERS

Electric charge

An Electric charge, denoted by Q is associated with a deficiency or the abundance of a group of electrons and is equal to the product of the number of electrons and the charge on each individual electron.

Electrical charge manifests itself in the form of forces—electrons repel other electrons but attract protons, while protons repel each other but attract electrons.

The unit of electrical charge is the coulomb(C). The **coulomb** is defined as the charge carried by 6.24×10^{18} electrons. Thus, if an electrically neutral (i.e., uncharged) body has 6.24×10^{18} electrons removed, it will be left with a net positive charge of 1 coulomb, i.e., $Q = 1 \text{ C}$. Conversely, if an uncharged body has 6.24×10^{18} electrons added, it will have a net negative charge of 1 coulomb, i.e., $Q = -1 \text{ C}$.

We can now determine the charge on one electron. It is $Q_e = 1 / (6.24 \times 10^{18}) = 1.60 \times 10^{-19} \text{ C}$.

Example - An initially neutral body has $1.7 \mu\text{C}$ of negative charge removed. Later, 18.7×10^{11} electrons are added. What is the body's final charge?

Solution - Initially the body is neutral, i.e., $Q_{\text{total}} = 0 \text{ C}$. When $1.7 \mu\text{C}$ of electrons is removed, the body is left with a positive charge of $1.7 \mu\text{C}$. Now, 18.7×10^{11} electrons are added back. This is equivalent to

$$18.7 \times 10^{11} \text{ electrons} \times \frac{1 \text{ coulomb}}{6.24 \times 10^{18} \text{ electrons}} = 0.3 \mu\text{C}$$

of negative charge. The final charge on the body is therefore $Q_f = 1.7 \mu\text{C} - 0.3 \mu\text{C} = 1.4 \mu\text{C}$.

Coulombs law

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Rachel S Tattersall



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