

Ideal gas law - problems and solutions

1. Ideal gases in a closed container initially have volume V and temperature T . The final temperature is $5/4T$ and the final pressure is $2P$. What is the final volume of gases.

Known :

Initial volume (V_1) = V

Initial temperature (T_1) = T

Final temperature (T_2) = $5/4 T$

Initial pressure (P_1) = P

Final pressure (P_2) = $2P$

Wanted : Final volume (V_2)

Solution :

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

$$\frac{P V}{T} = \frac{(2P) V_2}{5/4 T}$$

$$\frac{V}{1} = \frac{2 V_2}{5/4}$$

$$5/4 V = 2 V_2$$

$$V_2 = \frac{5V}{4} \cdot \frac{2}{1}$$

$$V_2 = \frac{5V}{4} \times \frac{1}{2}$$

$$V_2 = \frac{5V}{8}$$

2. Determine the volume of 2.00 moles of gases (ideal gas) at STP. STP = Standard Temperature and Pressure.

Known :

Moles of gas (n) = 2 moles

Standard temperature (T) = $0^\circ\text{C} = 0 + 273 = 273$ Kelvin

Standard pressure (P) = 1 atm = 1.013×10^5 Pa

Universal gas constant (R) = 8.315 Joule/mole.Kelvin

Wanted : Volume of gases (V)

Solution :

Read more :

<https://physics.gurumuda.net/ideal-gas-law-problems-and-solutions.htm>

Gas Law Problems With Solutions

Joseph Knox



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Dr. Salm's Chemistry Problem Drill Book Robert J. Salm, Robert J. Salm Robert J. Salm Ph.D., 2005 This drill book contains many common problem types that are asked in General Chemistry classes in High School and College This work will give you practice with the major problem types as you prepare for finals and standardized tests *Survival Guide to General Chemistry* Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields Journal of Chemical Education, 1925 Includes Report of New England Association of Chemistry Teachers and Proceedings of the Pacific Southwest Association of Chemistry Teachers Math Concepts for Food Engineering Richard W. Hartel, D.B. Hyslop, Robin K. Connelly, T.A. Howell Jr., 2008-03-17 A Supplement for Food Science Engineering Students Who Need to Improve Their Mathematical Skills A remedial textbook for understanding mathematical theories and formulas Math Concepts for Food Engineering Second Edition helps students improve their mathematical skills so that they can succeed in food engineering cour **Physico-chemical Calculations** Joseph Knox, 1919 **Questions and Answers on Federal Tax Laws, Problems Affecting Individuals, Partnerships and Corporations, January, 1924** Irving Bank-Columbia Trust Company, 1924 **University High School Journal**, 1925 **Student Solutions Manual for Chang's Chemistry** Philip C. Keller, Raymond Chang, Jill L. Keller, 1988 Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change *Physico-chemical Calculations* Joseph Knox, 1944 Chemical Symbolism and Calculations Stanley Wallace Morse, 1953 **Chemical Calculations** Raymond Harman Ashley, 1929 **Journal** University High School (Oakland, Calif.), 1926 **The Journal of Physical Chemistry**, 1912 *The Journal of Physical Chemistry* Wilder Dwight Bancroft, 1912 Includes section New Books Nature Sir Norman Lockyer, 1917 **Introduction to the Study of Chemical Philosophy** Sir

William Augustus Tilden, 1895 Journal of the American Chemical Society American Chemical Society, 1921 Proceedings of the Society are included in v 1 59 1879 1937 **Chemical News and Journal of Industrial Science**, 1908 Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry

The book delves into Gas Law Problems With Solutions. Gas Law Problems With Solutions is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Gas Law Problems With Solutions, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Gas Law Problems With Solutions
 - Chapter 2: Essential Elements of Gas Law Problems With Solutions
 - Chapter 3: Gas Law Problems With Solutions in Everyday Life
 - Chapter 4: Gas Law Problems With Solutions in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, the author will provide an overview of Gas Law Problems With Solutions. This chapter will explore what Gas Law Problems With Solutions is, why Gas Law Problems With Solutions is vital, and how to effectively learn about Gas Law Problems With Solutions.
3. In chapter 2, the author will delve into the foundational concepts of Gas Law Problems With Solutions. The second chapter will elucidate the essential principles that need to be understood to grasp Gas Law Problems With Solutions in its entirety.
4. In chapter 3, the author will examine the practical applications of Gas Law Problems With Solutions in daily life. This chapter will showcase real-world examples of how Gas Law Problems With Solutions can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Gas Law Problems With Solutions in specific contexts. The fourth chapter will explore how Gas Law Problems With Solutions is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Gas Law Problems With Solutions. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Gas Law Problems With Solutions.

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