



IONIC AND METALLIC BONDING

Chapter Test A

A. Matching

Match each definition with the correct term. Write the letter of the correct term in the space provided.

Column I

- _____ 1. electrical energy
- _____ 2. ionization
- _____ 3. lattice energy
- _____ 4. electron
- _____ 5. conductivity
- _____ 6. lattice ion
- _____ 7. atom
- _____ 8. compound
- _____ 9. ionization
- _____ 10. conductivity
- _____ 11. crystal lattice

Column II

- _____ 12. the energy released when one mole of gaseous ions forms one mole of solid ionic compound
- _____ 13. the energy required to separate one mole of a compound into its constituent ions
- _____ 14. the arrangement of ions in a crystal lattice
- _____ 15. a substance that conducts electricity
- _____ 16. a regular, repeating arrangement of ions in a crystal lattice
- _____ 17. energy that is used to break one mole of a compound into its constituent ions
- _____ 18. the energy required to separate one mole of a compound into its constituent ions
- _____ 19. the energy required to separate one mole of a compound into its constituent ions
- _____ 20. the energy required to separate one mole of a compound into its constituent ions

B. Multiple Choice

Write the letter of the correct answer in the space provided.

- _____ 21. How many electrons does a chlorine atom have?
 - (A) 17
 - (B) 18
 - (C) 19
 - (D) 20

Chapter 7 Ionic And Metallic Bonding Test Answers

Julie R. Frentrup



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