

Thermodynamic Equilibrium - 1 (Application of Equilibrium)

Thermodynamic Equilibrium

Thermodynamic Equilibrium states that both conditions are necessary to produce the equilibrium:

1. The temperature of system and surroundings (environment) are equal (thermal contact)
- Equal ✓
 - None
 - All of the above
2. The two thermodynamic states have the same pressure (mechanical contact)
- Equal ✓
 - None
 - All of the above
3. System and surroundings are doing the same physical work (pressure-volume work) (mechanical contact)
- None
 - Pressure
 - Temperature ✓
 - None of the above
4. Thermodynamic states (system and surroundings) are equal (mechanical contact)
- None
 - Temperature
 - Equal ✓
 - None of the above
5. Thermodynamic states (system and surroundings) are equal (mechanical contact)
- Equal temperature of system
 - Equal and homogeneous ✓
 - Equal force
 - Equal temperature of surroundings
6. Thermodynamic states (system and surroundings) are homogeneous
- Equal ✓
 - None
 - Equal
 - None
7. Thermodynamic states (system and surroundings) are equal (mechanical contact)
- Equal
 - None
 - Temperature ✓
 - None of the above
8. Thermodynamic states (system and surroundings) are equal (mechanical contact)
- Equal and homogeneous
 - Equal and homogeneous ✓
 - None of the above
 - None of the above

Chapter Test Thermal Energy Answers

TD Snyder



Chapter Test Thermal Energy Answers:

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