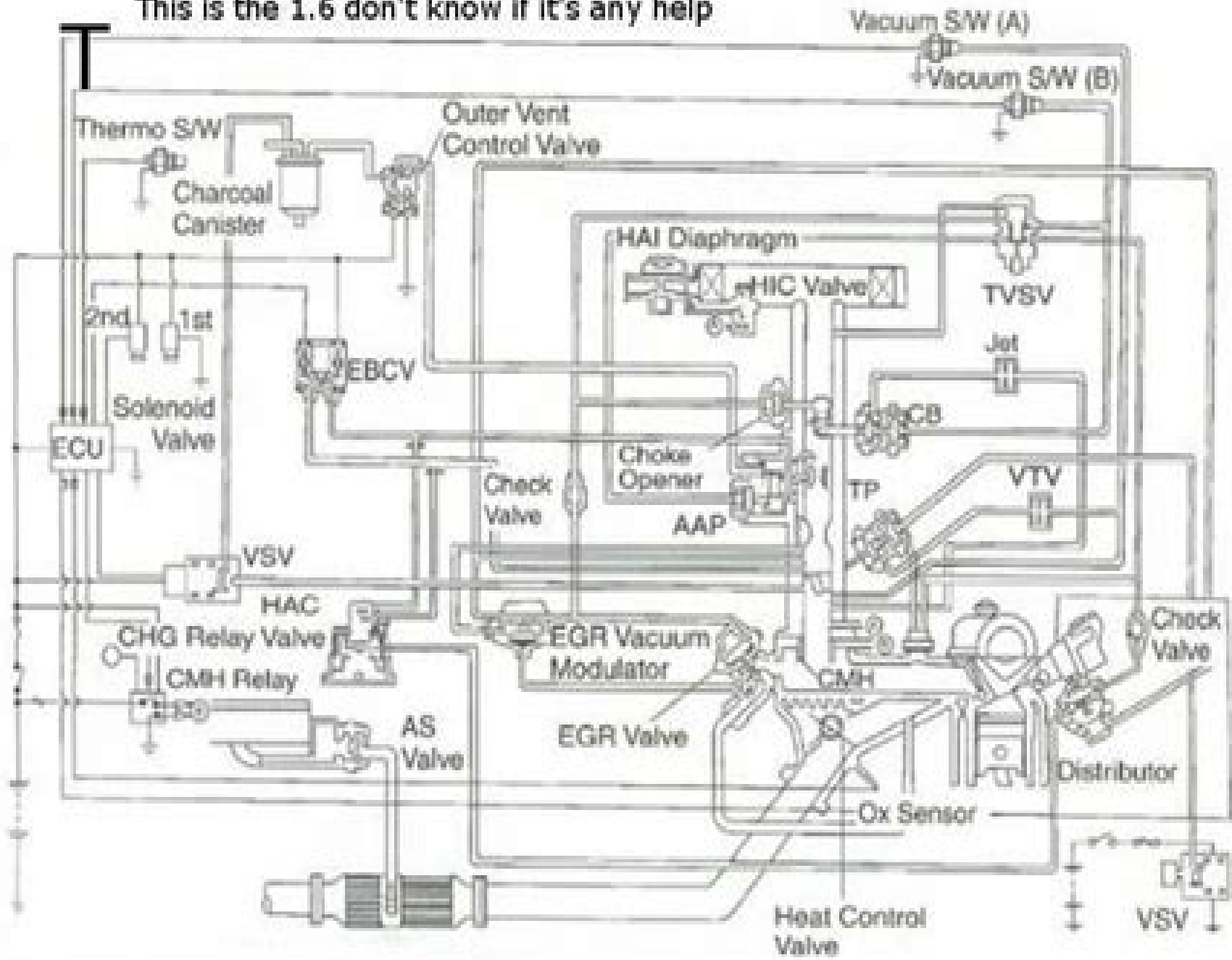


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# 2e Engine Wiring

**E Durkheim**



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web in the context of a chemical reaction the term neutralization is used for a reaction between an acid and a base or alkali historically this reaction was represented as acid base alkali salt water for example  $\text{hcl naoh nacl h}_2\text{o}$

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web during this reaction the  $\text{h}^+$  ions from the acid and the  $\text{oh}^-$  ions from the alkali react together to form water as this happens energy is released which is detected by the thermometer

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together to produce one mole of water for example  $\text{hcl aq naoh aq nacl aq h}_2\text{o aq}$  you can see that one mole of water is produced when one mole of hcl or one mole of naoh reacts

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