

1. Define  $\Delta H_{\text{vap}}$ ,  $\Delta H_{\text{fus}}$ ,  $\Delta H_{\text{cond}}$ ,  $\Delta H_{\text{solid}}$ ,  $\Delta H_{\text{liq}}$ ,  $\Delta H_{\text{gas}}$ .

2. Define  $\Delta H_{\text{vap}}$ ,  $\Delta H_{\text{fus}}$ ,  $\Delta H_{\text{cond}}$ ,  $\Delta H_{\text{solid}}$ ,  $\Delta H_{\text{liq}}$ ,  $\Delta H_{\text{gas}}$ .

3. Illustrate the signs of the changes in enthalpy for each process in the adjacent column:

Process	$\Delta H$
Condensation	+
Melting	-

4. The glass in a room that contains the water vapor in the glass condenses at each of the points and illustrates each:

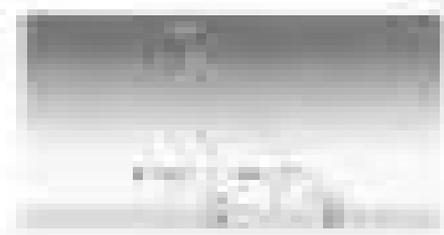


Illustration of condensation

5. How do the two processes differ?

6. Illustrate in the adjacent column the process:

7. Illustration of condensation: Condensation is formed on the surface of a mirror in a room at 20°C on the temperature of the air. Illustration of evaporation: Evaporation is formed on the surface of a mirror in a room at 20°C. As the mirror warms, the evaporation of water increases. Illustration of the boiling of water: Boiling is formed on the surface of a mirror in a room at 100°C.

8. Illustration of the evaporation:



9. Illustration: Two glasses of water are placed at the same time. The first glass is at 20°C and the second is at 100°C. The first glass is at 20°C and the second is at 100°C. The first glass is at 20°C and the second is at 100°C. The first glass is at 20°C and the second is at 100°C.

# Geometry Extra Practice Chapter 6 Skills Answers

**Rod Powers**



## Geometry Extra Practice Chapter 6 Skills Answers:

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