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**GEOGRAPHY**

**MAPWORK**

**NOTES**

**GRADES 10-12**

**Section 1**  
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web for example take the reaction of lead ii nitrate with hydrochloric acid to form lead ii chloride and nitric acid shown below  

$$\text{Pb(NO}_3)_2(\text{aq}) + 2\text{HCl}(\text{aq}) \rightarrow \text{PbCl}_2(\text{s}) + 2\text{HNO}_3$$

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web the net ionic equation for equation 1 is given as equation 3  $\text{s}_2 \text{aq}$   $\text{fe}_2 \text{aq}$   $\text{gt}$   $\text{fes}$   $\text{s}$   $\text{eq}$  3 8 note that the phase indicator  $\text{aq}$  is optional on ionic species but is

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web  $\text{cu clo}_4$  2 is soluble all perchlorates are soluble while  $\text{pbso}_4$  is insoluble sulfates are generally soluble except for  $\text{pb}^2$   $\text{ba}^2$   $\text{hg}_2^2$  and  $\text{sr}^2$  the net ionic equation is

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