

## Assessment

## Chapter Test A

## Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
  - a slippery feel
  - the ability to conduct an electric current
  - reactivity with metals
- \_\_\_\_\_ 2. Which is an example of a weak acid?
- HCl
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_3\text{PO}_4$
  - $\text{HNO}_3$
- \_\_\_\_\_ 3. Acetic acid is found in
- vinegar.
  - the stomach.
  - antacids.
  - oranges.
- \_\_\_\_\_ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
  - tastes bitter.
  - dissociates to form  $\text{OH}^-$  ions.
  - All of the above
- \_\_\_\_\_ 5. Which of the following is a Brønsted-Lowry base?
- HCl
  - $\text{HCO}_3^-$
  - $\text{H}_3\text{O}^+$
  - $\text{H}_3\text{PO}_4$
- \_\_\_\_\_ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(\text{aq}) + \text{CH}_3\text{NH}_2(\text{aq}) \rightleftharpoons \text{HC}_2\text{O}_4^-(\text{aq}) + \text{CH}_3\text{NH}_3^+(\text{aq}),$$
- which of these is a conjugate acid-base pair?
- $\text{H}_2\text{C}_2\text{O}_4(\text{aq})$  and  $\text{CH}_3\text{NH}_2(\text{aq})$
  - $\text{H}_2\text{C}_2\text{O}_4(\text{aq})$  and  $\text{CH}_3\text{NH}_3^+(\text{aq})$
  - $\text{CH}_3\text{NH}_2(\text{aq})$  and  $\text{CH}_3\text{NH}_3^+(\text{aq})$
  - $\text{HC}_2\text{O}_4^-(\text{aq})$  and  $\text{CH}_3\text{NH}_3^+(\text{aq})$

# Modern Chemistry Chapter Test

**Sabine Zange**

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## **Table of Contents Modern Chemistry Chapter Test**

1. Understanding the eBook Modern Chemistry Chapter Test
  - The Rise of Digital Reading Modern Chemistry Chapter Test
  - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Chapter Test
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Modern Chemistry Chapter Test
  - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Chapter Test
  - Personalized Recommendations
  - Modern Chemistry Chapter Test User Reviews and Ratings
  - Modern Chemistry Chapter Test and Bestseller Lists
5. Accessing Modern Chemistry Chapter Test Free and Paid eBooks
  - Modern Chemistry Chapter Test Public Domain eBooks

- Modern Chemistry Chapter Test eBook Subscription Services
- Modern Chemistry Chapter Test Budget-Friendly Options
- 6. Navigating Modern Chemistry Chapter Test eBook Formats
  - ePub, PDF, MOBI, and More
  - Modern Chemistry Chapter Test Compatibility with Devices
  - Modern Chemistry Chapter Test Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Modern Chemistry Chapter Test
  - Highlighting and Note-Taking Modern Chemistry Chapter Test
  - Interactive Elements Modern Chemistry Chapter Test
- 8. Staying Engaged with Modern Chemistry Chapter Test
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Modern Chemistry Chapter Test
- 9. Balancing eBooks and Physical Books Modern Chemistry Chapter Test
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Modern Chemistry Chapter Test
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Modern Chemistry Chapter Test
  - Setting Reading Goals Modern Chemistry Chapter Test
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modern Chemistry Chapter Test
  - Fact-Checking eBook Content of Modern Chemistry Chapter Test
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks

## 14. Embracing eBook Trends

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- Interactive and Gamified eBooks

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