

Important Questions for Class 9  
Science

Chapter 3 – Atoms and Molecules

Very Short Answer Questions

1 Mark

1. Atomic radius is measured in nanometers and

- (a)  $1\text{nm} = 10^{-10}\text{m}$
- (b)  $1\text{m} = 10^{-10}\text{nm}$
- (c)  $1\text{nm} = 10^{-9}\text{m}$
- (d)  $1\text{m} = 10^{-9}\text{nm}$

Ans: (c)  $1\text{nm} = 10^{-9}\text{m}$

2. Symbol of Iron is –

- (a) Ir
- (b) I
- (c) Fe
- (d) None of these

Ans: (c) Fe

3. Atomicity of Chlorine and Argon is

- (a) Diatomic and Monoatomic
- (b) Monoatomic and Diatomic
- (c) Monoatomic and Monoatomic
- (d) Diatomic and Diatomic

Ans: (a) Diatomic and Monoatomic

4. Molecular mass of water ( $\text{H}_2\text{O}$ ) is

- (a) 18g
- (b) 8g
- (c) 33g
- (d) 34g

Ans: (a) 18g

5. It is said that 1 mole of a compound contains –

- (a)  $6.023 \times 10^{23}$  atoms
- (b)  $6.023 \times 10^{28}$  atoms
- (c)  $60.23 \times 10^{23}$  atoms

# Chapter 3 Atoms And Moles

**PT Brinkman**



## **Chapter 3 Atoms And Moles:**

## Reviewing **Chapter 3 Atoms And Moles**: Unlocking the Spellbinding Force of Linguistics

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