

CHEM 101 Quiz #6 Chapter 6 ANSWERS**1. (4 points)**

The specific heat (s) of iron (Fe) is 0.40 J/(g·K). Calculate the heat effect q when 20 g of iron is heated from 100 °C to 200 °C. Include the sign!

$$q = m \cdot s \cdot \Delta T \text{ for } \Delta T \text{ in } \Delta K = \Delta ^\circ C \quad q = 20 \times 0.40 \times (200 - 100) = +800 \text{ J}$$

You must include the units and sign in the answer!

2. (4 points)

The Molar Mass of Mg = 24.0 g/mol.

Calculate the enthalpy change when 2.40 g Mg(s) reacts with excess oxygen. Include the sign!

$$2.40 \text{ g Mg} = 2.40/24.0 = 0.1 \text{ mol} \rightarrow \Delta H = 0.1(\text{mol}) \times (-600)(\text{kJ/mol}) = -60 \text{ kJ}$$

Again, your answer must include both the sign and the units.

3. (2 points) Which one of the following four reactions is a *formation reaction*? Circle the correct answer.



Look up the definition of a formation reaction in Chang or in your lecture notes: the compound is formed from its elements, and the elements must be as they naturally occur so e.g. Cl_2 not Cl and H_2 not H !

CHEM 101 Section L52 Quiz #6**1. (4 points)**

The specific heat (s) of copper (Cu) is 0.38 J/(g·K). Calculate the heat effect q when 10 g of copper is heated from 100 °C to 300 °C. Include the sign!

$$q = m \cdot s \cdot \Delta T \text{ for } \Delta T \text{ in } \Delta K = \Delta ^\circ C \quad q = 10 \times 0.38 \times (300 - 100) = +760 \text{ J}$$

You must include the units and sign in the answer!

2. (4 points)

The Molar Mass of Ca = 40.0 g/mol.

Calculate the enthalpy change ΔH when 8.0 g Ca(s) reacts with excess oxygen. Include the sign!!

$$8.0 \text{ g Ca} = 8.0/40 = 0.2 \text{ mol} \rightarrow \Delta H = 0.2(\text{mol}) \times (-650)(\text{kJ/mol}) = -130 \text{ kJ}$$

Again, your answer must include both the sign and the units.

3. (2 points) Which one of the following reactions is a *formation reaction*? Circle the correct answer.



Chem Fax Lab 6 Answers

Didier Musso



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Illinois Chemistry Teacher, 1993 **El-Hi Textbooks and Serials in Print, 1996** R R Bowker
Publishing, Bowker, 1996-02 Annual Register University of Chicago, 1909 Director's Report Cooper Union for the
Advancement of Science and Art, 1915 The President's Report Cooper Union for the Advancement of Science and
Art, 1916 **School Science and Mathematics**, 1918 **Bulletin - American Railway Engineering Association**
American Railway Engineering Association, 1930 Vols for 19 include the directory issue of the American Railway Engineering
Association **Journal of the American Chemical Society** American Chemical Society, 1926 **The Civil service**
competitor, 1892 Reports from Commissioners Great Britain. Parliament. House of Commons, 1875 ... **Catalogue of**
the Officers and Students of Leland University Leland College (Baker, La.), 1924 **Applied Spectroscopy**, 1963
Announcements for the Year ... Purdue University. Schools of Engineering, 1952 **An Introduction to the**
Practice of Organic Chemistry in the Laboratory Homer Adkins, Samuel Marion McElvain, Michael William Klein, 1940
Chemical Engineering Computing American Institute of Chemical Engineers, 1972 **Soviet Journal of Coordination**
Chemistry, 1978 *Documents of the Senate of the State of New York* New York (State). Legislature. Senate, 1902
Annual Report University of the State of New York. College Department, 1902 Annual Report University of the State
of New York. College Dept, 1902 **Professional Examination Papers**, 1901

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