

### Honors Chemistry - Chapter 12 Test

#### True/False - Clearly indicate the correct choice

1. In an aqueous solution of table salt (NaCl), the table salt would be considered the solvent.  
False
2. Water and gasoline are two liquids that would be considered miscible.  
False
3. A negative sign listed for the enthalpy of solution means that energy is released in the formation of the solution.  
True
4. A chemist finds that increasing the temperature leads to less of a substance dissolving in water. The substance is most likely a gas.  
True
5. Salt or sugar dissolved in water is considered to be a homogeneous mixture.  
True

#### Matching Set - Clearly indicate the correct choice

- a. solution    b. colloid    c. suspension
6. largest solute particle size c
  7. cannot be separated by filtration, will not scatter light a
  8. heterogeneous, does not separate on standing c
  9. particles settle out a
  10. smallest solute particle size b
  11. muddy water c
  12. shaving cream b

#### Multiple Choice - Clearly indicate the correct choice

13. A solute dissolved in alcohol is known as a/an  
a. aqueous solution    b. heterogeneous mixture    c. tincture  
d. immiscible solution  
c
14. An electrolyte is a substance that dissolves in water to form a solution

# Chemistry Chapter 12 Test Answers

**M Carnoy**



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## **Table of Contents Chemistry Chapter 12 Test Answers**

1. Understanding the eBook Chemistry Chapter 12 Test Answers
  - The Rise of Digital Reading Chemistry Chapter 12 Test Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Chapter 12 Test Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Chapter 12 Test Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Chapter 12 Test Answers
  - Personalized Recommendations
  - Chemistry Chapter 12 Test Answers User Reviews and Ratings
  - Chemistry Chapter 12 Test Answers and Bestseller Lists
5. Accessing Chemistry Chapter 12 Test Answers Free and Paid eBooks
  - Chemistry Chapter 12 Test Answers Public Domain eBooks
  - Chemistry Chapter 12 Test Answers eBook Subscription Services
  - Chemistry Chapter 12 Test Answers Budget-Friendly Options

6. Navigating Chemistry Chapter 12 Test Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Chapter 12 Test Answers Compatibility with Devices
  - Chemistry Chapter 12 Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Chapter 12 Test Answers
  - Highlighting and Note-Taking Chemistry Chapter 12 Test Answers
  - Interactive Elements Chemistry Chapter 12 Test Answers
8. Staying Engaged with Chemistry Chapter 12 Test Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Chapter 12 Test Answers
9. Balancing eBooks and Physical Books Chemistry Chapter 12 Test Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Chapter 12 Test Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Chapter 12 Test Answers
  - Setting Reading Goals Chemistry Chapter 12 Test Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Chapter 12 Test Answers
  - Fact-Checking eBook Content of Chemistry Chapter 12 Test Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

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