

**GENERAL CHEMISTRY-I (1411)**  
**S.L.# 5**

1. Define atomic number: \_\_\_\_\_  
Define mass number: \_\_\_\_\_  
Which of these can vary without changing the identity of an element?
- 
- 

2. How many protons, neutrons, and electrons are in the following atoms:

a. $^{40}\text{Ar}$	_____ p,	_____ n,	_____ e
b. $^{65}\text{Zn}$	_____ p,	_____ n,	_____ e
c. $^{75}\text{Ga}$	_____ p,	_____ n,	_____ e
d. $^{80}\text{Br}$	_____ p,	_____ n,	_____ e
e. $^{184}\text{W}$	_____ p,	_____ n,	_____ e
f. $^{241}\text{Am}$	_____ p,	_____ n,	_____ e

3. Locate each of the following elements in the periodic table; indicate whether it is a metal, metalloid, or nonmetal and give the name of the element.

a. Na	Name: _____	Class: _____
b. Ti	Name: _____	Class: _____
c. Ga	Name: _____	Class: _____
d. U	Name: _____	Class: _____
e. Pd	Name: _____	Class: _____
f. Se	Name: _____	Class: _____
g. Kr	Name: _____	Class: _____
h. As	Name: _____	Class: _____

4. Predict the charges of the ions of the following elements:

a. Mg	_____
b. Al	_____
c. K	_____
d. S	_____
e. F	_____
f. Sr	_____

# Chemistry The Central Science 11th Edition Answer Key

**D Keegan**



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