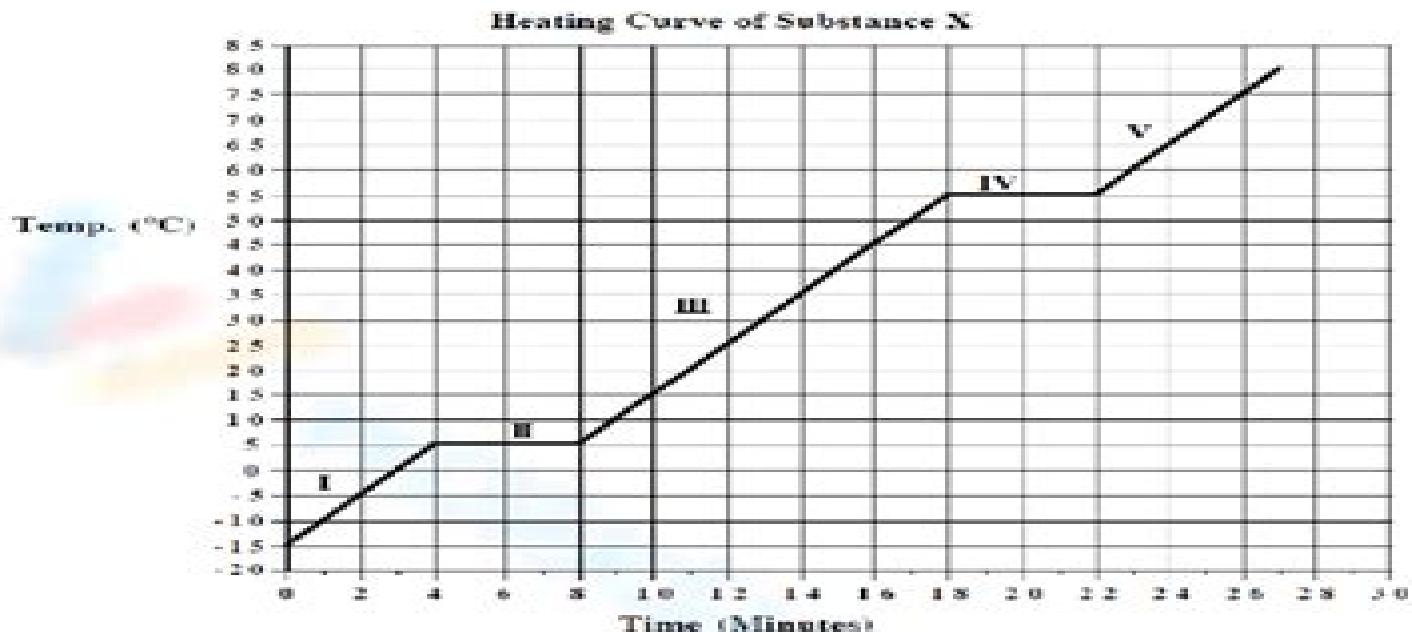


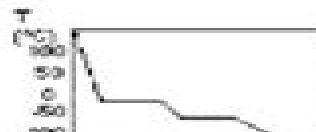
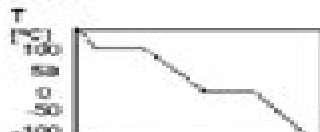
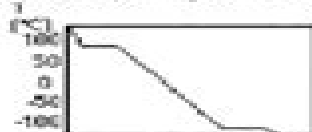
## Heating Curves Worksheet



The heating curve shown above is a plot of temperature vs time. It represents the heating of substance X at a constant rate of heat transfer. Answer the following questions using this heating curve:

- \_\_\_\_\_ 1. In what part of the curve would substance X have a definite shape and definite volume?
- \_\_\_\_\_ 2. In what part of the curve would substance X have a definite volume but no definite shape?
- \_\_\_\_\_ 3. In what part of the curve would substance X have no definite shape or volume?
- \_\_\_\_\_ 4. What part of the curve represents a mixed solid/liquid phase of substance X?
- \_\_\_\_\_ 5. What part of the curve represents a mixed liquid/vapor phase of substance X?
- \_\_\_\_\_ 6. What is the melting temperature of substance X?
- \_\_\_\_\_ 7. What is the boiling temperature of substance X?

Circle the correct cooling curve for water.



# Heating Curve Worksheet 1 Answers

**G Thomas**



## Heating Curve Worksheet 1 Answers:

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*Holt Science & Technology Calculator-Based Labs* Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004

**Introductory Chemical Engineering Thermodynamics** J. Richard Elliott,Carl T. Lira,2012-02-06 A Practical Up to Date Introduction to Applied Thermodynamics Including Coverage of Process Simulation Models and an Introduction to Biological Systems Introductory Chemical Engineering Thermodynamics Second Edition helps readers master the fundamentals of applied thermodynamics as practiced today with extensive development of molecular perspectives that enables adaptation to fields including biological systems environmental applications and nanotechnology This text is distinctive in making molecular perspectives accessible at the introductory level and connecting properties with practical implications Features of the second edition include Hierarchical instruction with increasing levels of detail Content requiring deeper levels of theory is clearly delineated in separate sections and chapters Early introduction to the overall perspective of composite systems like distillation columns reactive processes and biological systems Learning objectives problem solving strategies for energy balances and phase equilibria chapter summaries and important equations for every chapter Extensive practical examples especially coverage of non ideal mixtures which include water contamination via hydrocarbons polymer blending recycling oxygenated fuels hydrogen bonding osmotic pressure electrolyte solutions zwitterions and biological molecules and other contemporary issues Supporting software in formats for both MATLAB and spreadsheets Online supplemental sections and resources including instructor slides ConcepTests coursecast videos and other useful resources

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