

# Chemistry Worksheet BUNDLE



## Calculating Molar Mass

**Instructions:**  
Calculate the molar mass for each of the compounds in brackets. Write down the formula and include your units.



Example:  $\text{H}_2\text{O}$

(a)  $\text{H}_2\text{SO}_4$  (b)  $\text{CaCl}_2$  (c)  $\text{Al}_2\text{O}_3$  (d)  $\text{Fe}_2\text{O}_3$  (e)  $\text{CuSO}_4$

Compound	Formula	Molar Mass
Hydrogen peroxide	$\text{H}_2\text{O}_2$	
Ammonia	$\text{NH}_3$	
Hydrochloric acid	$\text{HCl}$	
Sulfuric acid	$\text{H}_2\text{SO}_4$	
Calcium hydroxide	$\text{Ca(OH)}_2$	
Sulfur dioxide	$\text{SO}_2$	
Carbon dioxide	$\text{CO}_2$	
Hydrogen sulfide	$\text{H}_2\text{S}$	
Acetic acid	$\text{CH}_3\text{COOH}$	
Glucose	$\text{C}_6\text{H}_{12}\text{O}_6$	
Urea	$\text{CO(NH}_2)_2$	
Ammonium sulfate	$(\text{NH}_4)_2\text{SO}_4$	
Calcium nitrate	$\text{Ca(NO}_3)_2$	
Silver chloride	$\text{AgCl}$	
Lead(II) nitrate	$\text{Pb(NO}_3)_2$	
Aluminum hydroxide	$\text{Al(OH)}_3$	
Iron(III) hydroxide	$\text{Fe(OH)}_3$	
Iron(II) hydroxide	$\text{Fe(OH)}_2$	
Iron(III) sulfate	$\text{Fe}_2(\text{SO}_4)_3$	
Iron(II) sulfate	$\text{FeSO}_4$	
Iron(III) chloride	$\text{FeCl}_3$	
Iron(II) chloride	$\text{FeCl}_2$	
Iron(III) oxide	$\text{Fe}_2\text{O}_3$	
Iron(II) oxide	$\text{FeO}$	
Iron(III) hydroxide	$\text{Fe(OH)}_3$	
Iron(II) hydroxide	$\text{Fe(OH)}_2$	
Iron(III) sulfate	$\text{Fe}_2(\text{SO}_4)_3$	
Iron(II) sulfate	$\text{FeSO}_4$	
Iron(III) chloride	$\text{FeCl}_3$	
Iron(II) chloride	$\text{FeCl}_2$	
Iron(III) oxide	$\text{Fe}_2\text{O}_3$	
Iron(II) oxide	$\text{FeO}$	

**ANSWER SHEETS!**



## Percent Composition

**Instructions:**  
Calculate the percent composition of each element in the compounds shown below. Show your work clearly.



Example:  $\text{H}_2\text{O}$

(a)  $\text{H}_2\text{O}$  (b)  $\text{H}_2\text{SO}_4$  (c)  $\text{CaCl}_2$  (d)  $\text{Al}_2\text{O}_3$  (e)  $\text{Fe}_2\text{O}_3$

Compound	Element	Percent
$\text{H}_2\text{O}$	H	
	O	
$\text{H}_2\text{SO}_4$	H	
	S	
	O	
	%	
$\text{CaCl}_2$	Ca	
	Cl	
$\text{Al}_2\text{O}_3$	Al	
	O	
	%	
$\text{Fe}_2\text{O}_3$	Fe	
	O	
	%	

**25 PROBLEMS**

# Calculating Molar Mass and Percent Composition



# Molar Mass And Percent Composition Worksheet Answers

**H Kauffman**



## **Molar Mass And Percent Composition Worksheet Answers:**

Chemistry Carson-Dellosa Publishing,2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

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**Chemistry Homework** Frank Schaffer Publications,Joan DiStasio,1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more

**Counting Moles** Nigel P. Freestone,2016-01-04 Students studying chemistry often struggle with the mole Counting Moles provides an effective aid to learning by giving clear and confident presentation of the essentials of the mole concept needed by those starting chemistry courses This user friendly self teach e book is split into six chapters which sequentially introduce the mole calculating frame to help solve problems Over 200 fully worked examples are given along with several hundred questions The mole concept is applied to topics such as relative atomic mass and relative formula mass percentage composition empirical and molecular formula The book also covers concentration its units volumetric analysis and the relationship between volume mass and moles of gases Counting Moles culminates in you taking a Mole Driving Test On passing this test you are issued with a Counting Moles Driving License that will give you all the confidence required to correctly answer all mole calculations

*Bite-Sized Chemistry Calculations* John Lambert,2024-07-10 Bite Sized Chemistry Calculations is a series of books on chemistry calculations aimed at helping students overcome the challenges associated with tackling the various types of calculations encountered in different aspects of chemistry focusing on a few topics at a time to facilitate comprehension Written by an experienced chemistry educator each book in the series has been tailored to fully meet the needs of students at all levels especially those taking college level general chemistry courses as well as those

following various O level curricula worldwide This part of the series explores the different types of problems and calculations encountered in mass the mole and stoichiometry including the determination of formulae of ionic compounds relative formula masses mass and percent compositions of compounds all aspects of mole calculations empirical and molecular formulae calculations based on chemical equations limiting reagents gas stoichiometry and percent yield The series is packed with many salient features that are meant to facilitate both teaching and learning Some of these include helpful explanations many examples alternative ways to solve problems plenty of practice questions complete answers an appendices With this book you will be well prepared for your exams and boost your performance

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*Formulas, Moles and Reactions* Alessio Mangoni, 2025-05-28 This comprehensive and rigorous educational text is dedicated to the study and application of the fundamentals of quantitative chemistry Through sixty clearly and precisely solved exercises it addresses essential topics such as molar mass and molecular weight the concept of the mole Avogadro s number the calculation of atoms and molecules in known quantities percent composition of compounds determination of empirical and molecular formulas and the balancing of chemical equations It serves as an effective tool for chemistry students and for anyone wishing to deepen their understanding of the discipline through a practical and methodical approach providing solid skills that are essential for continuing in scientific studies

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