

Periodic Table and Trends

The Periodic Table: Elements are arranged on the periodic table based on their physical and chemical properties. It was Dmitri Mendeleev (1834-1907) who first organized a table listing the elements in order of atomic mass in vertical columns. Each of the columns ended when the chemical properties of the elements started to repeat themselves. Atoms are listed in increasing atomic number and generally in increasing atomic mass.

Part A

Using colours and/or shading, indicate the following:

- Label Hydrogen
- Location of the metals, metalloids, non-metals
- The boundary between the metals and non-metals
- The following families: alkali metals, alkaline earth metals, transition metals, halogens, noble gases, lanthanides (rare earth metals), and actinides
- Label the groups I to VIII

Be sure to include a key, if necessary.

Handwritten Labels:

- Hydrogen (yellow)
- Alkali earth (light blue)
- transition metals (light purple)
- Basic Metal (arrow pointing to Al)
- Metalloids (green)
- Noble gases (light green)
- Alkali metal (pink)
- Non-metals (orange)
- Halogen (purple)

1 H	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Li	Be	B	C	N	O	F	Ne	Na	Mg	Al	Si	P	S	Cl	Ar	K	Ca
Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	Rb	Sr
Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	Cs	Ba
Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	Fr	Ra
Lr	Rf	Db	Sg	Bh	Hs	Mt	Uun	Uuu	Uub	Uuq	Uuq	Uuq	Uuq	Uuq	Uuq	Uuq	Uuq
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb				
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No				

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