

MODULE - 9 LECTURE NOTES - 4

DECISION SUPPORT SYSTEMS

INTRODUCTION

DSS is an interactive computer-based system to help decision makers use communications technologies, data, documents, knowledge and/or models to identify and solve problems, complete decision process tasks, and make decisions. Geographic Information Systems (GIS), Enterprise Information Systems (EIS), Expert Systems (ES), On-Line Analytical Processing (OLAP), software agents, knowledge discovery systems and group DSS can all be lumped into the category of systems we call as DSS.

DSS FOR DECISION-MAKING

Decision Preparation

- DSS environments provide data required as input to the decision-making process.
- This is what almost all data mart and data warehousing environments do today.

Decision structuring

- DSS environments provide tools and models for arranging the inputs in ways that make sense to frame the decision.
- These tools and models are not pivot tables and other aspects of data presentation found in query tools.
- They are actual decision-making tools, like fault tree analysis, Bayesian logic and model-based decision-making based on things like neural networks.

Context Development

- DSS environments again provide tools, and provide the mechanisms for capturing information about a decision's constituencies (who are affected by this decision), outcomes and their probabilities, and other elements of the larger decision-making context.

Decision-making

- DSS environments may automate all or part of the decision-making process and offer evaluations on the optimal decision.
- Expert systems and artificial intelligence environments can do this

Decision Propagation

- DSS environments take the information gathered about constituencies, dependencies, outcomes and drive elements of the decision into those constituencies for action.

Module 9 Lecture Notes Decision Support Systems

Introduction

Demetris Demetriou



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web log H⁺ = 10^{-4.05} = 8.91 × 10⁻⁵ mol dm⁻³ log pK_a = pH = 3.75 = 4.05 - 0.30 = 10 - 0.30 = 0.50 and HCOOH = 0.1 so HCOO⁻ = 0.200 mol dm⁻³ amount vol x

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hcl hydrochloric acid hno₃ nitric acid ch

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