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# Biology 101 Exam 1 With 100% Correct Answers 2023

covalent bond - Correct Answer-The sharing of a pair of valence electrons by two atoms. (Ex: Hydrogen, oxygen, water, Methane)

Polar covalent bond - Correct Answer-When electrons are not equally shared among the atoms within a bond. (Ex: water) This causes a negative and a positive side of a molecule.

Hydrogen Bond - Correct Answer-A type of weak chemical bond formed when the slightly positive hydrogen atom of a polar covalent bond in one molecule is attracted to the slightly negative atom of a polar covalent bond in another molecule.

Acid - Correct Answer-a solution with high amounts of H<sup>+</sup> (pH 0-6)

Base - Correct Answer-a solution with low amounts of H<sup>+</sup> (pH 8-14)

Neutral solution - Correct Answer-a solution with equal parts H<sup>+</sup> and OH<sup>-</sup>; pure water (pH ~ 7)

Buffer - Correct Answer-a substance that minimizes changes in pH. (Carbonic acid)

Hydroxyl group - Correct Answer-(-OH) can form hydrogen bonds with water , helps dissolve sugars

POLAR POLAR

Carboxyl group - Correct Answer-(-COOH) organic acid, found in acids i.e. vinegar. is

POLAR POLAR

Amino Group - Correct Answer-basic, can pick up and H<sup>+</sup>, In an amino acid

POLAR POLAR

Sulfhydryl group (-SH) - Correct Answer-(-SH) Resemble hydroxyl, POLAR POLAR

Phosphate group - Correct Answer-Backbone for phospholipids, most prevalent molecules in cell membranes

ATP - Correct Answer-adenosine triphosphate - 3 phosphate groups plus an adenosine, used for energy within cells. contains a ribose sugar.

# Biology 101 Test And Answers

**L Cohen**



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