

Molarity Practice Worksheet

Find the molarity of the following solutions:

$$\frac{\text{mol solute}}{\text{L solution}}$$

- 1) 0.5 moles of sodium chloride is dissolved to make 0.05 liters of solution.

$$\frac{0.5 \text{ mol NaCl}}{0.05 \text{ L}} = 10 \text{ M} = 10 \frac{\text{mol}}{\text{L}}$$

- 2) 0.5 grams of sodium chloride is dissolved to make 0.05 liters of solution.

$$\frac{0.5 \text{ g NaCl}}{0.05 \text{ L}} \times \frac{1 \text{ mol NaCl}}{58.44 \text{ g NaCl}} = 0.17 \text{ M}$$

- 3) 0.5 grams of sodium chloride is dissolved to make 0.05 mL of solution.

$$\frac{0.5 \text{ g}}{58.44 \text{ g}} \times \frac{1 \text{ mol}}{1} = \frac{0.0085 \text{ mol}}{0.00005} = 170 \text{ M}$$

$0.05 \text{ mL} \times \frac{1 \text{ L}}{10^3 \text{ mL}}$

Molarity And Molality Practice Problems With Answers

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Molarity And Molality Practice Problems With Answers:

Molarity And Molality Practice Problems With Answers Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the power of words has been evident than ever. They have the capability to inspire, provoke, and ignite change. Such may be the essence of the book **Molarity And Molality Practice Problems With Answers**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall effect on readers.

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