

# IB MYP 4-5 Chemistry : International Union of Pure and Applied Chemistry Study Notes

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## Alkanes

- Alkanes are a type of hydrocarbon with single bonds and saturated.
- They all have a general molecular formula of  $C_n H_{2n+2}$
- In the structural formula keep in mind they have single bonds
- Empirical formula is not the simplest ratio.
- A list of the different Alkanes

**Table 18.1 The First Ten Straight-Chain Alkanes**

NAME	MOLECULAR FORMULA	STRUCTURAL FORMULA
methane	$CH_4$	$CH_4$
ethane	$C_2H_6$	$CH_3-CH_3$
propane	$C_3H_8$	$CH_3-CH_2-CH_3$
butane	$C_4H_{10}$	$CH_3-CH_2-CH_2-CH_3$
pentane	$C_5H_{12}$	$CH_3-CH_2-CH_2-CH_2-CH_3$
hexane	$C_6H_{14}$	$CH_3-CH_2-CH_2-CH_2-CH_2-CH_3$
heptane	$C_7H_{16}$	$CH_3-CH_2-CH_2-CH_2-CH_2-CH_2-CH_3$
octane	$C_8H_{18}$	$CH_3-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_3$
nonane	$C_9H_{20}$	$CH_3-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_3$
decane	$C_{10}H_{22}$	$CH_3-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_2-CH_3$

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**Annelies Wilder-Smith**



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