

**CHEMISTRY**

23. In 80 grams of NaOH, there are:

- A. 2 moles of NaOH
- B. 1 mole of NaOH
- C. 3 moles of NaOH
- D. 0.5 mole of NaOH

24.  $\alpha$  - rays (Alpha) are:

- A. fast moving electrons
- B. protons
- C. neutron
- D. positively charged helium nuclei

25. The oxidation number of all the elements in free state is:

- A. negative
- B. positive
- C. zero
- D. -1

26. Viscosity of liquids depends upon:

- A. Temperature
- B. Size of molecule
- C. Shape of molecule
- D. Inter - molecular forces
- E. All of the above

27. Which pair of gases have same rate of diffusion at STP?

- A.  $H_2$  and Ne
- B.  $O_2$  and CO
- C.  $CO_2$  and  $C_3H_8$
- D.  $CO_2$  and  $SO_2$

# Nts Test Sample Paper For Karachi University

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**Nts Test Sample Paper For Karachi University:**

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