

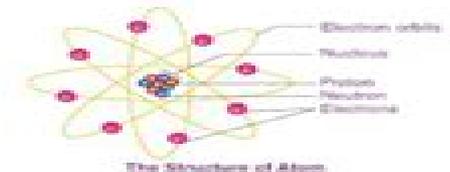
## Chapter 2: The Chemistry of Life

### ➤ Legos of the Universe

- Atoms
- Elements
  - Gold, carbon, silver, etc. (periodic table)

### ➤ Atomic Structure

- Nucleus
  - Proton (+)
  - Neutrons (no charge)
- Electron Shell



### ➤ Isotopes

- Atomic Mass

### ➤ Radioactive Atoms

- Some atomic nuclei are not stable and break down randomly
- These atoms are radioactive
- During decomposition they release, at a constant rate, a tiny high-speed particle carrying a lot of energy

### ➤ Electron shells determine bonding patterns

### ➤ Versatility of Carbon

- Can make 4 bonds
- Most often with C, H, O, and N
- Huge potential variety

### ➤ Ions

- Particles in the outer shell

### ➤ Changing Parts of the Atom

- What happens if you...
  - Add an electron to chlorine
  - Add a neutron to nitrogen
  - Add a proton to hydrogen



### ➤ Elements in Living Things

- Oxygen
- Carbon
- Hydrogen
- Nitrogen

### ➤ Molecules

- Molecules
- Bond energy
- Chemical reactions
  - **Reactants > Products**

### ➤ Chemical Bonding

- Covalent
- Ionic



# Chapter 2 The Chemistry Of Life Answer Key Concept Map

**Lakhmir Singh & Manjit Kaur**

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