

Important Questions for Class 9
Science

Chapter 3 – Atoms and Molecules

Very Short Answer Questions

1 Mark

1. Atomic radius is measured in nanometers and

- (a) $1\text{nm} = 10^{-10}\text{m}$
- (b) $1\text{m} = 10^{-10}\text{nm}$
- (c) $1\text{nm} = 10^{-9}\text{m}$
- (d) $1\text{m} = 10^{-9}\text{nm}$

Ans: (c) $1\text{nm} = 10^{-9}\text{m}$

2. Symbol of Iron is –

- (a) Ir
- (b) I
- (c) Fe
- (d) None of these

Ans: (c) Fe

3. Atomicity of Chlorine and Argon is

- (a) Diatomic and Monoatomic
- (b) Monoatomic and Diatomic
- (c) Monoatomic and Monoatomic
- (d) Diatomic and Diatomic

Ans: (a) Diatomic and Monoatomic

4. Molecular mass of water (H_2O) is

- (a) 18g
- (b) 8g
- (c) 33g
- (d) 34g

Ans: (a) 18g

5. It is said that 1 mole of a compound contains –

- (a) 6.023×10^{23} atoms
- (b) 6.023×10^{24} atoms
- (c) 60.23×10^{23} atoms

Chapter 3 Atoms And Moles

Masterton



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Atoms and Photons and Quanta, Oh My! F Todd Baker,2015-06-01 This is the second book in the Ask the Physicist series The first book From Newton to Einstein Ask the physicist about mechanics and relativity provides an excellent foundation for this book that covers topics in modern physics The main emphasis of this volume is providing an accessible introduction to quantum physics atomic physics and nuclear physics to anyone with at least high school physics knowledge

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