

## Chapter 4 – Atomic Structure

**Short Answer:** Answer the following questions in complete sentences using science vocabulary. Make sure that your answers are complete and detailed.

1. A sample of calcium contains calcium-40, calcium-44, calcium-42, calcium-48, calcium-43, and calcium-46 atoms. Explain why these atoms have different mass numbers but must have the same atomic number.

*The atoms have different mass numbers because they have a different number of neutrons. The mass number is equaled to the number of protons plus neutrons.*

*The atoms must have the same atomic number because the atomic number is equaled to the number of protons. The number of protons is unique for each element.*

2. Complete the following table. Circle the name of the element(s) that are isotopes. Complete Bohr's Model diagrams for the element with eleven electrons and the element with an atomic number of 9.

Element Name	Element Symbol	Atomic Number	Mass Number	Protons	Neutrons	Electrons
Scandium	Sc	21	45	21	24	21
Fluorine	F	9	19	9	10	9
Sodium	Na	11	24	11	13	11
Answers Vary					82	

# Chapter 4 Atomic Structure Practice Problems Answers

**Darrell D. Ebbing, Susan M. Schelble**



## Chapter 4 Atomic Structure Practice Problems Answers:

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific, detailed, step-by-step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid-base reactions, which often persists through working with acid-base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields. MCAT General Chemistry Review, 3rd Edition The Princeton Review, 2016-01-05 Make sure you're studying with the most up-to-date prep materials. Look for the newest edition of this title. The Princeton Review MCAT General Chemistry Review 4th Edition ISBN 9780593516256 on sale November 2022. Publisher's Note: Products purchased from third-party sellers are not guaranteed by the publisher for quality or authenticity and may not include access to online tests or materials included with the original product. *McGraw-Hill's SAT Subject Test: Chemistry, 2ed* Thomas Evangelist, 2009-02-01 We want to help you score high on the SAT Chemistry test. We've put all of our proven expertise into McGraw-Hill's SAT Subject Test Chemistry to make sure you're fully prepared for this difficult exam. With this book, you'll get essential skill-building techniques and strategies created by a leading high school chemistry teacher. You'll also get 5 full-length practice tests, hundreds of sample questions, and all the facts about the current exam. With McGraw-Hill's SAT Subject Test Chemistry, we'll guide you step-by-step through your preparation program and give you the tools you need to succeed. 5 full-length sample tests with complete explanations for every question. 40 top test items to remember on exam day. A step-by-step review of all topics covered on the exam. Teacher-recommended tips and strategies to help you raise your score. *McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition* Thomas Evangelist, 2011-02-03 Expert guidance on the Chemistry exam. Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects. McGraw-Hill's SAT Subject Test Chemistry is written by experts in the field and gives you the guidance you need to perform at your best. This book includes 4 full-length sample tests updated for the latest test format. 40 top tips to remember on test day. Glossary of tested chemistry terms and formulas. Tips and strategies from one of

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Chapter 4 Atomic Structure Practice Problems Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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