

**CHAPTER 7 REVIEW***Chemical Formulas and Chemical Compounds***MIXED REVIEW**

**SHORT ANSWER** Answer the following questions in the space provided.

1. Write formulas for the following compounds:

- \_\_\_\_\_ a. copper(II) carbonate  
\_\_\_\_\_ b. sodium sulfite  
\_\_\_\_\_ c. ammonium phosphate  
\_\_\_\_\_ d. tin(IV) sulfide  
\_\_\_\_\_ e. nitrous acid

2. Write the Stock names for the following compounds:

- \_\_\_\_\_ a.  $\text{Mg}(\text{ClO}_4)_2$   
\_\_\_\_\_ b.  $\text{Fe}(\text{NO}_3)_2$   
\_\_\_\_\_ c.  $\text{Fe}(\text{NO}_2)_3$   
\_\_\_\_\_ d.  $\text{CoO}$   
\_\_\_\_\_ e. dinitrogen pentoxide

3. \_\_\_\_\_ a. How many atoms are represented by the formula  $\text{Ca}(\text{HSO}_4)_2$ ?  
\_\_\_\_\_ b. How many moles of oxygen atoms are in a 0.50 mol sample of this compound?  
\_\_\_\_\_ c. Assign the oxidation number to sulfur in the  $\text{HSO}_4^-$  anion.

4. Assign the oxidation number to the element specified in each of the following:

- \_\_\_\_\_ a. hydrogen in  $\text{H}_2\text{O}_2$   
\_\_\_\_\_ b. hydrogen in  $\text{MgH}_2$   
\_\_\_\_\_ c. sulfur in  $\text{S}_8$   
\_\_\_\_\_ d. carbon in  $(\text{CO}_3)^{2-}$   
\_\_\_\_\_ e. chromium in  $\text{Na}_2\text{Cr}_2\text{O}_7$   
\_\_\_\_\_ f. nitrogen in  $\text{NO}_2$

# Chapter 7 3 Answers Chemical Formulas And Chemical Compounds

**William S. Seese, Guido H. Daub**



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## **Table of Contents Chapter 7 3 Answers Chemical Formulas And Chemical Compounds**

1. Understanding the eBook Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - The Rise of Digital Reading Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Personalized Recommendations
  - Chapter 7 3 Answers Chemical Formulas And Chemical Compounds User Reviews and Ratings
  - Chapter 7 3 Answers Chemical Formulas And Chemical Compounds and Bestseller Lists
5. Accessing Chapter 7 3 Answers Chemical Formulas And Chemical Compounds Free and Paid eBooks
  - Chapter 7 3 Answers Chemical Formulas And Chemical Compounds Public Domain eBooks
  - Chapter 7 3 Answers Chemical Formulas And Chemical Compounds eBook Subscription Services
  - Chapter 7 3 Answers Chemical Formulas And Chemical Compounds Budget-Friendly Options

6. Navigating Chapter 7 3 Answers Chemical Formulas And Chemical Compounds eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 7 3 Answers Chemical Formulas And Chemical Compounds Compatibility with Devices
  - Chapter 7 3 Answers Chemical Formulas And Chemical Compounds Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Highlighting and Note-Taking Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Interactive Elements Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
8. Staying Engaged with Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
9. Balancing eBooks and Physical Books Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Setting Reading Goals Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Fact-Checking eBook Content of Chapter 7 3 Answers Chemical Formulas And Chemical Compounds
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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