

Chapter Test A

Chapter: Chemical Equations and Reactions

Do the work carefully, using the letter of the letter of phrases that best completes each statement or best answers each question.

1. The mass number of an atom is the sum of the number of _____ and the number of _____ in the nucleus of the atom.
 - a. electrons; neutrons
 - b. protons; neutrons
 - c. protons; electrons
 - d. neutrons; electrons
2. The number of _____ in a neutral atom is equal to the number of _____ in the atom.
 - a. electrons; protons
 - b. protons; neutrons
 - c. neutrons; electrons
 - d. electrons; neutrons
3. Which of the following is a chemical equation?
 - a. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - b. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O} + 2\text{H}_2\text{O}$
 - c. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 - d. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
4. Which of the following is a chemical equation?
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 - d. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
5. Which of the following is a chemical equation?
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 - b. The number of _____ in a neutral atom is equal to the number of _____ in the atom.
 - c. The number of _____ in a neutral atom is equal to the number of _____ in the atom.
 - d. The number of _____ in a neutral atom is equal to the number of _____ in the atom.
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 - c. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
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Chapter 9 Modern Chemistry Test

Merle M. Ohlsen



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Handbook of Modern Pharmaceutical Analysis Satinder Ahuja, Stephen Scypinski, 2001. This book describes the role modern pharmaceutical analysis plays in the development of new drugs. Detailed information is provided as to how the quality of drug products is assured from the point of discovery until the patient uses the drug. Coverage includes state-of-the-art topics such as analytics for combinatorial chemistry and high-throughput screening, formulation development, stability studies, international regulatory aspects, and documentation and future technologies that are likely to impact the field. Emphasis is placed on current, easy-to-follow methods that readers can apply in their laboratories. No book has effectively replaced the very popular text *Pharmaceutical Analysis* that was edited in the 1960s by Tak Higuchi. This book will fill that gap with an up-to-date treatment that is both handy and authoritative.

The Good Student Charles Jay Rop, 1996

Evaluation in Modern Education Jacob Wayne Wrightstone, 1956

Laboratory Experiments to Accompany "Modern Chemistry," Fredus Nelson Peters, 1902

Religion, Magic, and Science in Early Modern Europe and America Allison P. Coudert, 2011-10-17. This fascinating study looks at how the seemingly incompatible forces of science, magic, and religion came together in the 15th, 16th, and 17th centuries to form the foundations of modern culture. As *Religion, Magic, and Science in Early Modern Europe and America* makes clear, the early modern period was one of stark contrasts: witch burnings and the brilliant mathematical physics of Isaac Newton; John Locke's plea for tolerance and the palpable lack of it; the richness of intellectual and artistic life and the poverty of material existence for all but a tiny percentage of the population. Yet for all the poverty, insecurity, and superstition, the period produced a stunning galaxy of writers, artists, philosophers, and scientists. This book looks at the conditions that fomented the emergence of such outstanding talent, innovation, and invention in the period 1450 to 1800. It examines the interaction between religion, magic, and science during that time, the impossibility of clearly differentiating between the three, and the impact of these forces on the geniuses who laid the foundation for modern science and culture.

Elementary Modern Chemistry Wilhelm Ostwald, Harry Wheeler Morse, 1909

Modern Road Emulsions Road Emulsion Association. Technical Committee, 1958

Scientific Protocols for Fire Investigation John J. Lentini, 2006-01-26. *Scientific Protocols for Fire Investigation* provides comprehensive coverage from historical, developmental, current, and practical perspectives. The author, uniquely qualified with years of experience in both on-site investigations and lab analyses, provides a resource that is unparalleled in depth and focus. The book is distinctive in that it not

Modern Microcrystal Tests for Drugs Charles Clarke Fulton, 1969

Report of the New England Association of Chemistry Teachers ... New England Association of Chemistry Teachers, 1931

Guidance Services in the Modern School Merle M. Ohlsen, 1974

Introduction to Modern Chemistry: Student Manual Eugene Meyer, 1979

Chemistry in the Modern

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