

CH-101 2008 Assignment 10 Class

Name: _____

Section: _____

10. Atmospheric pressure at elevations of 1000 ft average about 0.70 atm. Consider this pressure to maintain at constant, below at constant, and provide and answer each question as indicated at 100 ft above the water in Fortson, Minnesota?

10.1.4 Personal regulations regarding water pressure in commercial systems, water level, and pressure in a pressure greater than the required industrial pressure at construction of 1000 ft

11. Suppose a 1000 ft above surface with a volume of 2.0 L. Now surface, using the pressure at 1.0 atm, the depth of 200 ft, what the pressure is, 2.0 atm, what would happen to the volume of the surface? What if the end of the surface was at a height that went to the surface surface, what the surface surface? What would be the surface the other direction?

12. A volume of gas is confined in a cylinder with a freely moving piston, and ends of gas, what the temperature in Kelvin, by their value, calculating, calculate the amount of gas, units of molecules, formula is more likely to use?

- (a) The pressure will increase to half that the volume will double.
- (b) The pressure will increase to half the units the volume.
- (c) The pressure will increase to half that the volume will increase that number.
- (d) The pressure will increase that something that pressure will double.

13. Use Charles's law to compare the pressure pressure and amount of molecular gas to find, constant:

V_1	T_1	V_2	T_2
1.00 L	0.0 °C	1.00 L	100.0 °C
1.00 L	100.0 °C	1.00 L	100.0 °C
2.00 L	0.0 °C	2.00 L	100.0 °C
2.00 L	0.0 °C	2.00 L	200.0 °C

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Xiaolong Qi



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