

# Writing Ionic Formulas

14

Directions: In each table below, correctly identify the element symbols, charges, and the chemical formula for each ionic compound listed. Once a table is completely correct, the table will turn green!

out of 14

Name:	1. Sodium Chloride			
Symbols:	Na		Cl	
Charges:	+1	-	-1	-
Formula:	NaCl			

Name:	2. Calcium Fluoride			
Symbols:	Ca		F	
Charges:	+2	-	-1	-
Formula:	CaF <sub>2</sub>			

Name:	3. Magnesium Phosphide			
Symbols:	Mg		P	
Charges:	+2	-	-3	-
Formula:	Mg <sub>3</sub> P <sub>2</sub>			

Name:	4. Beryllium Iodide			
Symbols:	Be		I	
Charges:	+2	-	-1	-
Formula:	BeI <sub>2</sub>			

Name:	5. Sodium Oxide			
Symbols:	Na		O	
Charges:	+1	-	-2	-
Formula:	Na <sub>2</sub> O			

Name:	6. Aluminum Sulfide			
Symbols:	Al		S	
Charges:	+3	-	-2	-
Formula:	Al <sub>2</sub> S <sub>3</sub>			

Name:	7. Copper (I) Bromide			
Symbols:	Cu		Br	
Charges:	+1	-	-1	-
Formula:	CuBr			

Name:	8. Sodium Nitrate			
Symbols:	Na		NO <sub>3</sub>	
Charges:	+1	-	-1	-
Formula:	NaNO <sub>3</sub>			

Name:	9. Potassium Sulfate			
Symbols:	K		SO <sub>4</sub>	
Charges:	+1	-	-2	-
Formula:	K <sub>2</sub> SO <sub>4</sub>			

Name:	10. Ammonium Hydroxide			
Symbols:	NH <sub>4</sub>		OH	
Charges:	+1	-	-1	-
Formula:	NH <sub>4</sub> OH			

# Chemfax Ionic Formula Writing Kit Answers

**Sebastian Brünink**



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