

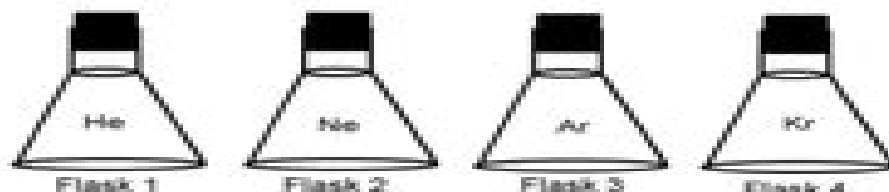
Gases Test SCH3U

Name: _____

 K T C A **Assessment of Learning (Product)**
15 9 5 10

Multiple Choice. Circle the most correct answer. (15K)

1. Convert the pressure 1.80 kPa to atm.
A) 2 atm B) 115 atm C) 132 atm D) 245 atm E) None of the above
2. What happens to the temperature of a gas when it is expanded quickly?
A) Temperature increases B) Temperature does not change C) Temperature decreases



3. If all of the following flasks are the same size, at the same temperature, and contain the same number of molecules, in which flask will the pressure be highest?
A) All have the same pressure B) Flask 1 C) Flask 2
D) Flask 3 E) Flask 4
4. Which diagram best represents a gas in a closed container?



5. According to the kinetic molecular theory, the Kelvin temperature of an ideal gas is proportional to which one of the following.
A) acceleration B) momentum C) kinetic energy D) potential energy
6. A gas is _____ and assumes _____ of its container whereas a liquid is _____ and assumes _____ of its container.
A) compressible, the volume and shape, not compressible, the shape of a portion
B) compressible, the shape, not compressible, the volume and shape
C) compressible, the volume and shape, compressible, the volume
D) condensed, the volume and shape, condensed, the volume and shape
E) condensed, the shape, compressible, the volume and shape

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CH Cherryholmes



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Table of Contents Chemistry Gases Unit Study Answers

1. Understanding the eBook Chemistry Gases Unit Study Answers
 - The Rise of Digital Reading Chemistry Gases Unit Study Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Gases Unit Study Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Gases Unit Study Answers
 - User-Friendly Interface

4. Exploring eBook Recommendations from Chemistry Gases Unit Study Answers
 - Personalized Recommendations
 - Chemistry Gases Unit Study Answers User Reviews and Ratings
 - Chemistry Gases Unit Study Answers and Bestseller Lists
5. Accessing Chemistry Gases Unit Study Answers Free and Paid eBooks
 - Chemistry Gases Unit Study Answers Public Domain eBooks
 - Chemistry Gases Unit Study Answers eBook Subscription Services
 - Chemistry Gases Unit Study Answers Budget-Friendly Options
6. Navigating Chemistry Gases Unit Study Answers eBook Formats
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 - Chemistry Gases Unit Study Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Gases Unit Study Answers
 - Highlighting and Note-Taking Chemistry Gases Unit Study Answers
 - Interactive Elements Chemistry Gases Unit Study Answers
8. Staying Engaged with Chemistry Gases Unit Study Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Gases Unit Study Answers
9. Balancing eBooks and Physical Books Chemistry Gases Unit Study Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Gases Unit Study Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Gases Unit Study Answers
 - Setting Reading Goals Chemistry Gases Unit Study Answers
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Chemistry Gases Unit Study Answers
 - Fact-Checking eBook Content of Chemistry Gases Unit Study Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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