

Chapter 7: Solutions and Colloids

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CHAPTER OUTLINE

7.1 Physical States of Solutions	7.4 Solution Concentrations	7.7 Solution Properties
7.2 Solubility	7.5 Solution Preparation	7.8 Colloids
7.3 The Solution Process	7.6 Solution Stoichiometry	7.9 Dialysis

LEARNING OBJECTIVES/ASSESSMENT

When you have completed your study of this chapter, you should be able to:

1. Classify mixtures as solutions or nonsolutions based on their appearance. (Section 7.1; Exercise 7.4)
2. Demonstrate your understanding of terms related to the solubility of solutes in solution. (Section 7.2; Exercises 7.6 and 7.12)
3. Predict in a general way the solubilities of solutes in solvents on the basis of molecular polarity. (Section 7.3; Exercise 7.16)
4. Calculate solution concentrations in units of molarity, weight/weight percent, weight/volume percent, and volume/volume percent. (Section 7.4; Exercises 7.22 b, 7.30 c, 7.34 a, and 7.38 c)
5. Describe how to prepare solutions of specific concentration using pure solutes and solvent, or solutions of greater concentration than the one desired. (Section 7.5; Exercises 7.46 and 7.48 b)
6. Do stoichiometric calculations based on solution concentrations. (Section 7.6; Exercise 7.56)

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Chemistry for Today: General, Organic, and Biochemistry Spencer L. Seager, Michael R. Slabaugh, 2013-01-01

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provides greater support in both problem solving and critical thinking skills By demonstrating how this information will be important to a reader s future career and providing important career information online the authors not only help readers to set goals but also to focus on achieving them

Evolving Nature of Objectivity in the History of Science and its Implications for Science Education Mansoor Niaz,2017-10-26 This book explores the evolving nature of objectivity in the history of science and its implications for science education It is generally considered that objectivity certainty truth universality the scientific method and the accumulation of experimental data characterize both science and science education Such universal values associated with science may be challenged while studying controversies in their original historical context The scientific enterprise is not characterized by objectivity or the scientific method but rather controversies alternative interpretations of data ambiguity and uncertainty Although objectivity is not synonymous with truth or certainty it has eclipsed other epistemic virtues and to be objective is often used as a synonym for scientific Recent scholarship in history and philosophy of science has shown that it is not the experimental data Baconian orgy of quantification but rather the diversity plurality in a scientific discipline that contributes toward understanding objectivity History of science shows that objectivity and subjectivity can be considered as the two poles of a continuum and this dualism leads to a conflict in understanding the evolving nature of objectivity The history of objectivity is nothing less than the history of science itself and the evolving and varying forms of objectivity does not mean that one replaced the other in a sequence but rather each form supplements the others This book is remarkable for its insistence that the philosophy of science and in particular that discipline s analysis of objectivity as the supposed hallmark of the scientific method is of direct value to teachers of science Meticulously yet in a most readable way Mansoor Niaz looks at the way objectivity has been dealt with over the years in influential educational journals and in textbooks it s fascinating how certain perspectives fade while basic questions show no sign of going away There are few books that take both philosophy and education seriously this one does Roald Hoffmann Cornell University chemist writer and Nobel Laureate in Chemistry

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