

## Chemistry Final 2013 – Semester 1 Review Guide

### I. CLASSIFICATION OF MATTER

1. Define the following:

- a. Elements: \_\_\_\_\_
- b. Compound: \_\_\_\_\_
- c. Mixture: \_\_\_\_\_

2. What is the difference between homogeneous and heterogeneous mixtures?

3. Identify the following as an element (e), compound (c), or mixture (m):

| Substance    | E, C or M |
|--------------|-----------|
| As           |           |
| NaCl         |           |
| Sand + Water |           |
| S            |           |
| Air          |           |

4. Which of the following indicate a chemical change:

- (a) Fizzing  
(b) Burning  
(c) Color change  
(d) All of the above

5. Which of the following are physical changes:

- (a) State of changes (solid, liquid, gas)  
(b) Dissolving  
(c) Cracking open an egg  
(d) All of the above

6. What is the difference between a physical change and a chemical change?

### II. LABORATORY SAFETY

1. When should long hair be tied back?

2. When chemicals are spilled what should you do?

# Chemistry Midterm Exam Review

**Rachel Sandford**



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