

Section 12.1

Section 12.1

Answer complete answer multiple-choice questions that are provided.

1. _____
 - a. _____
 - (a) correct, in terms of all the reactants and products.
 - (b) correct, in terms of moles of reactants and products.
 - (c) correct, in terms of work done by work done by the system.
 - (d) correct, in terms of enthalpy change, ΔH .
 - b. _____
 - (a) the heat capacity of the system.
 - (b) the mass of the system.
 - (c) the number of moles of hydrogen that will react with a certain quantity of oxygen.
 - (d) the initial enthalpy of the system.
 - c. _____
 - (a) the heat capacity of the system.
 - (b) the mass of the system.
 - (c) the number of moles of hydrogen that will react with a certain quantity of oxygen.
 - (d) the initial enthalpy of the system.
 - d. _____
 - (a) the heat capacity of the system.
 - (b) the mass of the system.
 - (c) the number of moles of hydrogen that will react with a certain quantity of oxygen.
 - (d) the initial enthalpy of the system.
2. _____
 - a. _____
 - (a) correct, in terms of all the reactants and products.
 - (b) correct, in terms of moles of reactants and products.
 - (c) correct, in terms of work done by work done by the system.
 - (d) correct, in terms of enthalpy change, ΔH .
 - b. _____
 - (a) the heat capacity of the system.
 - (b) the mass of the system.
 - (c) the number of moles of hydrogen that will react with a certain quantity of oxygen.
 - (d) the initial enthalpy of the system.
 - c. _____
 - (a) the heat capacity of the system.
 - (b) the mass of the system.
 - (c) the number of moles of hydrogen that will react with a certain quantity of oxygen.
 - (d) the initial enthalpy of the system.
 - d. _____
 - (a) the heat capacity of the system.
 - (b) the mass of the system.
 - (c) the number of moles of hydrogen that will react with a certain quantity of oxygen.
 - (d) the initial enthalpy of the system.

3. Answer the two questions based on the following balanced equation:



a. Balance the equation:



b. Write the molar mass of H_2 or O_2 .

c. How many moles of H_2 are needed to react with 1.0 mole of O_2 ?

d. How many moles of H_2O are formed from 1.0 mole of O_2 ?

e. How many grams of H_2 are needed to react with 1.0 mole of O_2 ?

Chapter 9 Section 1 Stoichiometry Answers

Ying Liu



Chapter 9 Section 1 Stoichiometry Answers:

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In some sort of inundated with screens and the cacophony of immediate communication, the profound power and mental resonance of verbal art usually fade in to obscurity, eclipsed by the constant onslaught of sound and distractions. Yet, set within the lyrical pages of **Chapter 9 Section 1 Stoichiometry Answers**, a fascinating work of fictional splendor that impulses with natural thoughts, lies an unique trip waiting to be embarked upon. Published with a virtuoso wordsmith, this exciting opus instructions readers on a psychological odyssey, softly revealing the latent possible and profound affect embedded within the complicated internet of language. Within the heart-wrenching expanse with this evocative analysis, we shall embark upon an introspective exploration of the book is key subjects, dissect its captivating publishing style, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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