

## Chemistry Chapter 8 Exam

### Multiple Choice

1. Which of the following is not a Lewis acid?

- \_\_\_\_\_ a.  $\text{CH}_3\text{COOH}$
- \_\_\_\_\_ b. Lewis Acids
- \_\_\_\_\_ c. Carbonyl Compounds
- \_\_\_\_\_ d.  $\text{BF}_3$
- \_\_\_\_\_ e. Lewis Bases
- \_\_\_\_\_ f.  $\text{Mg}^{2+}$
- \_\_\_\_\_ g. Lewis Acids
- \_\_\_\_\_ h.  $\text{H}^+$
- \_\_\_\_\_ i.  $\text{H}_2\text{O}$

- \_\_\_\_\_ j.  $\text{H}_2\text{O}$
- \_\_\_\_\_ k.  $\text{H}_2\text{O}$
- \_\_\_\_\_ l.  $\text{H}_2\text{O}$
- \_\_\_\_\_ m.  $\text{H}_2\text{O}$
- \_\_\_\_\_ n.  $\text{H}_2\text{O}$
- \_\_\_\_\_ o.  $\text{H}_2\text{O}$
- \_\_\_\_\_ p.  $\text{H}_2\text{O}$
- \_\_\_\_\_ q.  $\text{H}_2\text{O}$
- \_\_\_\_\_ r.  $\text{H}_2\text{O}$
- \_\_\_\_\_ s.  $\text{H}_2\text{O}$
- \_\_\_\_\_ t.  $\text{H}_2\text{O}$
- \_\_\_\_\_ u.  $\text{H}_2\text{O}$
- \_\_\_\_\_ v.  $\text{H}_2\text{O}$
- \_\_\_\_\_ w.  $\text{H}_2\text{O}$
- \_\_\_\_\_ x.  $\text{H}_2\text{O}$
- \_\_\_\_\_ y.  $\text{H}_2\text{O}$
- \_\_\_\_\_ z.  $\text{H}_2\text{O}$

### Multiple Choice

2. The strongest Lewis acid is

- \_\_\_\_\_ a.  $\text{H}^+$
- \_\_\_\_\_ b.  $\text{H}_2\text{O}$
- \_\_\_\_\_ c.  $\text{H}_2\text{O}$
- \_\_\_\_\_ d.  $\text{H}_2\text{O}$

3. The Lewis base,  $\text{H}_2\text{O}$ , reacts with  $\text{H}^+$  to form  $\text{H}_3\text{O}^+$ . The Lewis acid is  $\text{H}^+$ .

- \_\_\_\_\_ a.  $\text{H}^+$
- \_\_\_\_\_ b.  $\text{H}_2\text{O}$
- \_\_\_\_\_ c.  $\text{H}_2\text{O}$
- \_\_\_\_\_ d.  $\text{H}_2\text{O}$

4. The Lewis base,  $\text{H}_2\text{O}$ , reacts with  $\text{H}^+$  to form  $\text{H}_3\text{O}^+$ . The Lewis acid is  $\text{H}^+$ .

- \_\_\_\_\_ a.  $\text{H}^+$
- \_\_\_\_\_ b.  $\text{H}_2\text{O}$
- \_\_\_\_\_ c.  $\text{H}_2\text{O}$
- \_\_\_\_\_ d.  $\text{H}_2\text{O}$

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**Annelies Wilder-Smith**



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web he is a faculty member at istanbul university department of chemistry he received his undergraduate degree from kocaeli university department of chemistry and his graduate and doctorate degrees from istanbul university institute of science inorganic chemistry program his main research subjects are the design and synthesis of coordination

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