

# pH

Name \_\_\_\_\_

pH is a scale that measures the hydronium ion concentration of a solution. Therefore, the pH scale can be used to determine the acidity of a solution. A pH of less than 7 indicates an acidic solution, a pH of 7 is neutral, and a pH of greater than 7 up to 14 is basic. The lower the pH, the higher the acidity. The higher the pH, the lower the acidity.

Indicators are substances that change color at a different pH levels. Phenolphthalein is colorless in an acid and a neutral solution, pink in a base. Blue litmus changes to red in an acid, and remains blue in neutral and basic solutions. Red litmus remains red in acidic and neutral substances, but turns blue in bases.

Complete the following chart.

pH	Acid, Base, Neutral	Phenolphthalein	Blue Litmus	Red Litmus
2				
8				
4				
7				
13				
11				
5				
1				

# Ph Of Salt Solutions Physical Science If8767

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