

STOICHIOMETRY

QUESTIONS

1. Will an increase in temperature always increase the solubility of the solute in the solvent? Why or why not?

No, because in the case of solid solutes, the effect of the temperature towards solubility depends on what kind of reaction, either exothermic or endothermic. In endothermic reaction as the temperature increases, the solubility increases as well but in exothermic reaction, if the temperature increases, the solubility decreases. The same manner with gas particles, increased temperature, makes higher solubility.

2. Why is glycerine soluble in water but not in diethyl ether?

It is because water is a polar solvent making glycerine (a polar solute) to become soluble, however diethyl ether is a non-polar solvent making glycerine insoluble.

3. How will you construct a procedure in preparation of 3 M NaCl (100mL solution)?

First, compute for the needed grams/mol of NaCl to put in the solvent. Since we need to prepare a 3 M NaCl in 100ml solution, we compute it by $0.100L(3\text{mol})/1L=0.3 \text{ mol NaCl}$ then multiplying 0.3 mol NaCl to 58.43 g/mol NaCl (molar mass of NaCl compound) resulting to 17.529 g/NaCl ,which will be the amount in weight that will be put in the solvent. Second, measure as close to 17.529 g NaCl in weighing paper using a balance.

Then transfer it to 100 ml volumetric flask. Third, put the water solvent in the in the bottom of the flask in approximately half full then shake to dissolve the NaCl solute. Lastly, carefully put more water until reaching the rest of the bulb area up to the neck of the flask then add solvent drop-wise monitoring the location of the meniscus, then labeled the flask for laboratory protocols.

● CONCLUSION

Summarized the basic properties and preparation of solutions. Solutions are known to be the homogeneous mixture where in substances and its components are uniformly distributed. Solvents are components that are greater amounts and the solute which is the lesser amount that is miscible or soluble. Solubility is

Experiment 6 Stoichiometry Lab Report Conclusion

RS Peters



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The Student's Lab Companion John W. Lehman, 2008 This comprehensive lab companion provides enough theory to help students understand how and why an operation works but emphasizes the practical aspects of an operation to help them perform the operation successfully in the lab For undergraduate or graduate students taking organic chemistry lab This comprehensive lab companion provides enough theory to help students understand how and why an operation works but emphasizes the practical aspects of an operation to help them perform the operation successfully in the lab The Second Edition makes substantive revisions of many operations to clarify existing material and add new information More environmentally friendly i e green lab experiments are encouraged Ideal for professors who write their own lab experiments or would like custom labs but need a source for lab operations and safety information *Foundations of College Chemistry, Laboratory* Morris Hein, Susan Arena, 2010-08-09 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis *Laboratory Manual for Principles of General Chemistry* J. A. Beran, 2013-12-12 A two term manual for General Chemistry This supplementary manual focuses on chemical principles and techniques The Laboratory Manual for Principles of General Chemistry tenth edition provides a broad scope of experiments coupled with a clear layout for ease of use The manual delivers material for two or three course terms It also assists chemistry students in knowing how to time various techniques in the lab environment The companion manual is organized into topic sections such as Chemical and Physical Properties Atomic and Molecular Structure Gases and Solutions *Practical Biochemistry for Colleges* E. J. Wood, 2012-12-02 This book presents a selection of tried and trusted laboratory experiments in the field of biochemistry The experiments are described in detail and can be used directly or in a modified form They are grouped according to a broad range of biochemical disciplines which allows those responsible for arranging practical classes to select experiments to complement any given biochemistry course Suggestions are made for further work in more advanced classes As well as the practical method the experiments are accompanied by background information discussion of results references for further study and illustrations *Scientific and Technical Aerospace Reports* ,1968 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database [Using Multimedia Technology in Chemistry Pre-laboratory Preparation](#) Jeffrey Glen Yoder, 2002 **Rare Metal Technology 2022** Takanari Ouchi, Gisele Azimi, Kerstin Forsberg, Hojong Kim, Shafiq Alam, Neale R. Neelameggham, Alafara Abdullahi Baba, Hong Peng, 2022-02-02 This collection presents papers

from a symposium on extraction of rare metals from primary and secondary materials and residues as well as rare extraction processing techniques used in metal production Authors cover the extraction of less common or minor metals including elements such as antimony bismuth barium beryllium boron calcium chromium gallium germanium hafnium indium manganese molybdenum platinum group metals rare earth metals rhenium scandium selenium sodium strontium tantalum tellurium and tungsten Contributions also discuss rare metals of low tonnage sales compared to high tonnage metals iron copper nickel lead tin zinc or light metals such as aluminum magnesium or titanium and electronic metalloid silicon Authors also cover biometallurgy hydrometallurgy and electrometallurgy while novel high temperature processes such as microwave heating solar thermal reaction synthesis and cold crucible synthesis of rare metals are addressed Also included in this collection is the design of extraction equipment used in these processes from suppliers as well as laboratory and pilot plant studies

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A Selected Listing of NASA Scientific and Technical Reports États-Unis. National aeronautics and space administration. Scientific and technical information division,1965

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