

## Chemistry – Unit 4 Notes Dalton's Playbook

In the late 19<sup>th</sup> century, British physicist John Dalton developed and refined his atomic model. Dalton's model was based on experiments that showed Dalton's Atomic Theory. His theory was the first atomic model. The substance is made up of particles called atoms. Atoms cannot be created or destroyed. Atoms are indivisible particles that cannot be created or destroyed. Atoms are present in matter. Matter is made up of particles called atoms. Atoms are present in matter. Matter is made up of particles called atoms.

### Table 1 – Properties

Property	Atoms	Molecules	Elements
Can be created or destroyed			
Can be divided			

1. What happens when two atoms of the same element combine to form a molecule?
2. What happens when two different atoms of the same element and the same number of atoms combine?
3. What happens when two different atoms of the same element and the same number of atoms combine?

### Table 2 – Examples

Fill in the table by giving the correct name of the atoms and the number of atoms in each molecule.

Molecule	Number of atoms	Element	Atomic mass
Water (H <sub>2</sub> O)			
Carbon dioxide (CO <sub>2</sub> )			
Ammonia (NH <sub>3</sub> )			
Hydrogen chloride (HCl)			
Glucose (C <sub>6</sub> H <sub>12</sub> O <sub>6</sub> )			

1. Write the chemical formula for the following molecules: (a) Water (b) Carbon dioxide (c) Ammonia (d) Hydrogen chloride (e) Glucose
2. How many atoms are present in the following molecules? (a) Water (b) Carbon dioxide (c) Ammonia (d) Hydrogen chloride (e) Glucose

# Modeling Chemistry Dalton Playhouse Notes Answers

**Shasha Hu**



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