

Stoichiometry: Mass – Mole Worksheet

Solve the following stoichiometry problems and type in the correct answer with units.
 (for example: 3.45 mol or 4.62 g) Round the masses from the periodic table to the nearest whole number. Round all answers to 2 decimal places.

1. Based on the following chemical equation, please answer the following questions



- a. Calculate the amount of moles of NaCl needed to produce 75 g of FeCl₂.

Answer: _____ $\frac{\text{g FeCl}_2}{1 \text{ mol FeCl}_2} \times \frac{1 \text{ mol FeCl}_2}{\text{g FeCl}_2} \times \frac{\text{mol NaCl}}{\text{mol FeCl}_2} =$

- b. Calculate the amount of Fe₂(SO₄)₃ in grams produce 215 g Na₂SO₄.

Answer: _____ $\frac{\text{g Na}_2\text{SO}_4}{1 \text{ mol Na}_2\text{SO}_4} \times \frac{1 \text{ mol Na}_2\text{SO}_4}{\text{g Na}_2\text{SO}_4} \times \frac{\text{mol Fe}_2(\text{SO}_4)_3}{\text{mol Na}_2\text{SO}_4} \times \frac{\text{g Fe}_2(\text{SO}_4)_3}{1 \text{ mol Fe}_2(\text{SO}_4)_3} =$

- c. Calculate the amount in grams of FeCl₂ produced with 95g of Fe(SO₄)₂.

Answer: _____ $\frac{\text{g Fe}(\text{SO}_4)_2}{1 \text{ mol Fe}(\text{SO}_4)_2} \times \frac{1 \text{ mol Fe}(\text{SO}_4)_2}{\text{g Fe}(\text{SO}_4)_2} \times \frac{\text{mol FeCl}_2}{\text{mol Fe}(\text{SO}_4)_2} \times \frac{\text{g FeCl}_2}{1 \text{ mol FeCl}_2} =$

- d. Calculate the amount of moles of Fe₂(SO₄)₃ needed to react with 123.5 g of NaCl.

Answer: _____ $\frac{\text{g NaCl}}{1 \text{ mol NaCl}} \times \frac{1 \text{ mol NaCl}}{\text{g NaCl}} \times \frac{\text{mol Fe}_2(\text{SO}_4)_3}{\text{mol NaCl}} =$

2. How many moles of CaCO₃ can be produced if we make 12.3 g of Ca(HCO₃)₂ reacts with CaCO₃?



Answer: _____ $\frac{\text{g Ca}(\text{HCO}_3)_2}{1 \text{ mol Ca}(\text{HCO}_3)_2} \times \frac{1 \text{ mol Ca}(\text{HCO}_3)_2}{\text{g Ca}(\text{HCO}_3)_2} \times \frac{\text{mol CaCO}_3}{\text{mol Ca}(\text{HCO}_3)_2} =$

Moles And Stoichiometry Packet Answers

NARAYAN CHANGDER



Moles And Stoichiometry Packet Answers:

Moles And Stoichiometry Packet Answers: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous captivating novels enthraling the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the captivating narratives that have captivated audiences this year. Moles And Stoichiometry Packet Answers : Colleen Hoover's "It Ends with Us" This poignant tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Moles And Stoichiometry Packet Answers : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Moles And Stoichiometry Packet Answers : Delia Owens "Where the Crawdads Sing" This evocative coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is an exceptional and suspenseful novel that will keep you guessing until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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