

CHAPTER 6

CHAPTER ASSESSMENT

The Periodic Table and Periodic Law

Reviewing Vocabulary

Match the definition in Column A with the term in Column B.

Column A

Column B

Period

- Statement that when the elements are arranged by increasing atomic number, there is a periodic repetition of their chemical and physical properties
- Groups 1 and 2, 13 through 18
- Groups 3 through 12
- Group 1 elements (except for hydrogen)

Alkaline earth metals

5. Group 2 elements

- A column in the periodic table
- A row in the periodic table

halogens

- Group 17 elements
- Group 18 elements

- Atom or bonded group of atoms that has a positive or negative charge
- Energy required to remove an electron from a gaseous atom
- Statement that atoms tend to gain, lose, or share electrons to acquire a full set of eight valence electrons

electronegativity

- Indication of an atom's ability to attract electrons in a chemical bond

Write a sentence that uses each group of terms.

14. transition metals, inner transition metals

The fact the two best conductors of electricity are a transition metal, inner transition metals the fact the two _____

15. metal, nonmetal, metalloid

Chapter 6 The Periodic Table Law Assessment Answers

Rosina Ehmann



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