



IONIC AND METALLIC BONDING

Chapter Test A

A. Matching

Match each definition with the correct term. Write the letter of the correct term in the space provided.

Column I

- _____ 1. electrical energy
- _____ 2. ionization
- _____ 3. lattice energy
- _____ 4. electron
- _____ 5. conductivity
- _____ 6. lattice ion
- _____ 7. atom
- _____ 8. compound
- _____ 9. ionization
- _____ 10. conductivity
- _____ 11. crystal lattice

Column II

- _____ 12. the energy released when one mole of electrons is added to one mole of atoms
- _____ 13. the energy released when one mole of gaseous ions is formed
- _____ 14. the energy required to separate one mole of ions
- _____ 15. the smallest amount of energy to produce electrons
- _____ 16. a substance that conducts heat and electricity
- _____ 17. regular structure of atoms
- _____ 18. the energy required to separate one mole of atoms
- _____ 19. the energy required to separate one mole of ions
- _____ 20. the energy required to separate one mole of electrons
- _____ 21. the energy required to separate one mole of atoms

B. Multiple Choice

Write the letter of the correct answer in the space provided.

- _____ 22. How many electrons does a chlorine atom have?
 - (A) 17
 - (B) 18
 - (C) 19
 - (D) 20

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