

You must show all your calculation set-ups to receive full credit. You must report all answers to the correct # of significant figures (sig. figs.).

1. One mole of any gas at STP occupies a volume of

- a. 22.4 L  
 b. 1.00 L  
 c. 24.4 L  
 d. 24.4 mL  
 e. 1.00 mL

$$\frac{85}{100} + 9 = \frac{94}{100} \text{ great!}$$

2. The density of a gas is reported in units of

- a. g/cm<sup>3</sup>  
 b. atm  
 c. mm Hg  
 d. g/mL  
 e. g/L

3. Which color of visible light will have the longest wavelength of radiation?

- a. red  
 b. yellow  
 c. blue  
 d. violet  
 e. orange

VR to GR

4. For an exothermic reaction,  $\Delta H$  is always

- a. positive  
 b. negative  
 c. zero  
 d. small

$$\Delta H = \frac{-q}{n}$$

5. When calculating the root mean square velocity of a sample of chlorine gas, we should use which of the following for the molar mass?

- a. 35.45 g/mol  
 b. 0.03545 kg/mol  
 c. 70.90 g/mol  
 d. 0.07090 kg/mol

$$r_{ms} = \sqrt{\frac{3RT}{M}} \rightarrow \text{kg/mol } (35.45 \text{ g}) \frac{1 \text{ kg}}{1000 \text{ g}} = 0.03545 \text{ kg}$$

6. The symbol (\*) in  $\Delta H^*$  represents which of the following conditions?

- a. 1 atm, 0°C  $\rightarrow$   $\Delta P$   
 b. 1 atm, 298K  $\rightarrow$   $\Delta E$   
 c. 1.013 atm, 25°C  
 d. 1.013 atm, 273K

# Chem 1a Lab Manual Answers Fresno State

**ML Yell**



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