


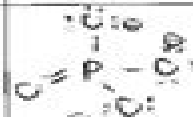




1. Determine the molecular geometry for the following molecules. Determine the hybridization of the central atom.

Compound	Lewis structure	# electron domains	Electronic geometry	Molecular geometry	Hybridization
SO <sub>2</sub>		3	trigonal planar	bent	sp <sup>2</sup>
NH <sub>3</sub>		4	tetrahedral	trigonal pyramidal	sp <sup>3</sup>
CH <sub>2</sub> O		3	trigonal planar	trigonal planar	sp <sup>2</sup>
PO <sub>4</sub> <sup>3-</sup>		4	tetrahedral	tetrahedral	sp <sup>3</sup>
C <sub>2</sub> H <sub>2</sub>		2	linear	linear	sp
I <sub>3</sub> <sup>-</sup>		5	trigonal bipyramidal	linear	sp <sup>3</sup> d

# Chemical Bonding Lab Answers

**William L. Masterton, Stanley M.  
Cherim**



## **Chemical Bonding Lab Answers:**

## Unveiling the Magic of Words: A Review of "**Chemical Bonding Lab Answers**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Chemical Bonding Lab Answers**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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