

Chemistry Final Exam Study Guide 2021-22

The final exam will cover chapters 6, 7, 8 and 9, with a few very basic questions from chapters 10 and 11. Recommended ways to prepare for final exam.

- **review objectives, notes, worksheets, charts, quizzes and exams.**
- **make your own study guide**
- if you missed questions on test or quizzes, reread those topics in the textbook
- practice problems from each chapter. Appendix D in Pearson Chemistry has a problem bank with many of the answers included in the appendix. Use these for practice
- use quizzes for reviewing vocabulary, naming compounds, or other topics, most chapters of Pearson have quizzes already made or you can make your own... I have reviewed the quizzes listed on your powerschool pages and listed only those I felt were accurate.
- Study buddies, quizzing classmates or other students taking chemistry
- Take the standardized test in the textbook at the end of each chapter. I will post the answers to each of these

The exam will consist of 65-80 questions, all multiple choice. There will be copies of the periodic table, polyatomic ion chart, and transition metal charts attached to the back of the exam. You will need your calculator to complete the final exam.

Chapter 6: The Periodic Table (about 15 questions)

Vocabulary

proton
electron
atomic number
period
group
metal
nonmetal
representative element (Group A)
transition metal (Group B)
halogens
noble gas

Objectives (see p. 185 6.1 and 6.2)

- Know how the periodic table is arranged.
- Be able to use the periodic table to locate alkali metals, Group A elements, Group B elements and periods.
- Differentiate between metals, metalloids, and nonmetals.
- Be able to explain why noble gases can exist as single atoms, and usually don't react with other elements.
- ***Know how to determine the number of valence electrons in a Group A element.**

Chemistry Final Exam Study Guide Multiple Choice

**Kathleen M. Savage, Charity Anne
Dorgan**



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Resources in Education, 1988 **Basic Concepts of Chemistry, Study Guide** Leo J. Malone, 2000 The text's three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems *Complete Book of Graduate Programs in the Arts and Sciences* Princeton Review (Firm), 2004-09 Our Best 357 Colleges is the best selling college guide on the market because it is the voice of the students Now we let graduate students speak for themselves too in these brand new guides for selecting the ideal business law medical or arts and humanities graduate school It includes detailed profiles rankings based on student surveys like those made popular by our Best 357 Colleges guide as well as student quotes about classes professors the social scene and more Plus we cover the ins and outs of admissions and financial aid Each guide also includes an index of all schools with the most pertinent facts such as contact information And we've topped it all off with our school says section where participating schools can talk back by providing their own profiles It's a whole new way to find the perfect match in a graduate school Study Guide for Introductory Chemistry Iris Stovall, 1990 Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Clinical Medicine; Examination Review** George Horemis, 1972
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